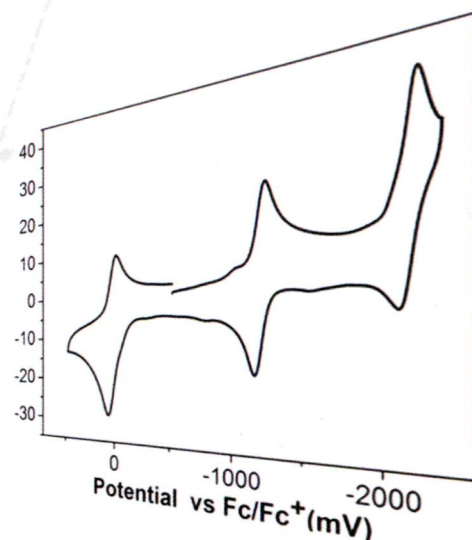
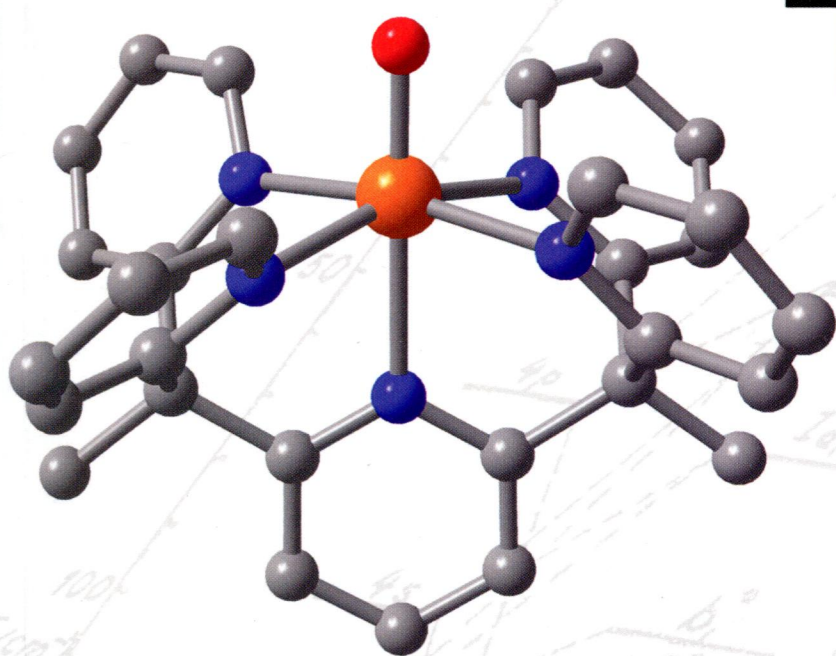


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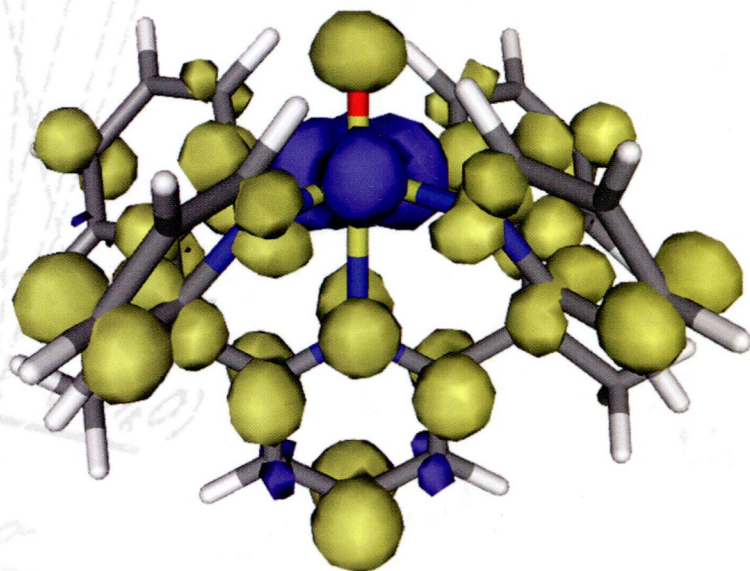
# Inorganic Chemistry

including bioinorganic chemistry

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Low-Valent  
Vanadium Oxo  
Chemistry



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**ON THE COVER:** The synthesis and characterization of a unique terminal mononuclear vanadium(III) oxo species supported by a pentadentate polypyridyl ligand are reported. Computational results predict that this  $[V^{III}O]^+$  species features the lowest possible oxidation state of vanadium in this ligand environment. The results have important implications for future efforts toward low-valent vanadyl chemistry, particularly with regard to the isolation and study of formal vanadium(II) oxo species. [The cover background is a digitally modified version of Figure 3 from the article "The Electronic Structure of the Vanadyl Ion" by C. J. Ballhausen and Harry B. Gray (*Inorg. Chem.* **1962**, *1*, 111–122).] See A. E. King, M. Nippe, M. Atanasov, T. Chantarojsiri, C. A. Wray, E. Bill, F. Neese, J. R. Long, and C. J. Chang, p 11388.

## Communications

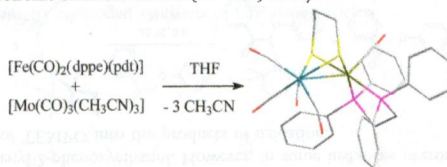
11345

DOI: 10.1021/ic501875q

### A New FeMo Complex as a Model of Heterobimetallic Assemblies in Natural Systems: Mössbauer and Density Functional Theory Investigations

Solène Bouchard, Martin Clémancey, Geneviève Blondin,\* Maurizio Bruschi,\* Kévin Charreureur, Luca De Gioia, Christine Le Roy, François Y. Pétillon, Philippe Schollhammer,\* and Jean Talarmin

The novel FeMo heterobimetallic species  $[FeMo(CO)_5(\kappa^2\text{-dppe})(\mu\text{-pdt})]$  was synthesized by reacting two mononuclear precursors,  $[Fe(CO)_2(\kappa^2\text{-dppe})(\kappa^2\text{-pdt})]$  and  $[Mo(CO)_3(CH_3CN)_3]$ , in order to obtain an original platform for the activation of resource molecules. Mössbauer spectroscopy and density functional theory calculations were performed to describe the proper representation of the electronic structure of this {Fe–Mo} entity.



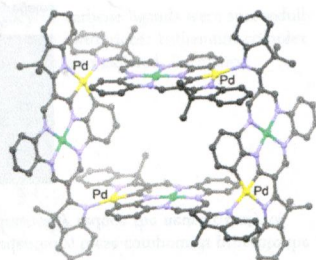
11348

DOI: 10.1021/ic5019828

### Heterometallic Pd<sup>II</sup>–Ni<sup>II</sup> Complexes with *meso*-Substituted Dibenzotetraaza[14]annulene: Double C–H Bond Activation and Formation of a Rectangular Tetradibenzotetraaza[14]annulene

Hamid Khaleedi,\* Marilyn M. Olmstead, Takamitsu Fukuda, and Hapipah Mohd Ali

C–H bond activation at palladium(II) centers led to the transformation of a cofacial bis(tetraaza[14]annulene) to a unique tetrapalladamacrocyclic rectangle.

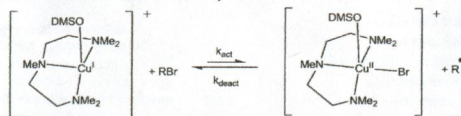




**New Method for Exploring Deactivation Kinetics in Copper-Catalyzed Atom-Transfer-Radical Reactions**

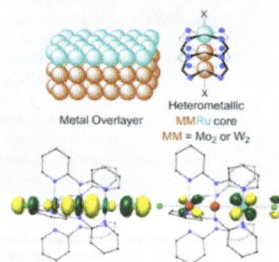
Timothy J. Zerk and Paul V. Bernhardt\*

Copper(I) complexes may react with an alkyl halide to release an alkyl radical (activation). The kinetics of the reverse reaction between the alkyl radical and copper(II) complex are less studied because it is difficult to isolate or quantify the concentration of the alkyl radical ( $R^\bullet$ ) in situ. We report a broadly applicable electrochemical technique for simultaneously measuring the kinetics of deactivation and kinetics of activation in these systems.

**Heterometallic Multiple Bonding: Delocalized Three-Center  $\sigma$  and  $\pi$  Bonding in Chains of 4d and 5d Transition Metals**

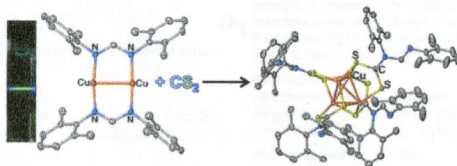
David W. Brogden and John F. Berry\*

$\text{Mo}_2\text{Ru}(\text{dpa})_4\text{Cl}_2$  (**1**) and  $\text{W}_2\text{Ru}(\text{dpa})_4\text{Cl}_2$  (**2**) have short Mo–Ru and W–Ru distances, 2.38 Å (**1**) and 2.39 Å (**2**), indicating delocalized Mo–Mo–Ru and W–W–Ru bonding. Density functional theory calculations reveal the presence of both a three-center/two-electron  $\sigma$  bond and two three-center/four-electron  $\pi$  bonds in the compounds.

**Articles****Multinuclear Copper(I) and Silver(I) Amidinate Complexes: Synthesis, Luminescence, and  $\text{CS}_2$  Insertion Reactivity**

Andrew C. Lane, Matthew V. Vollmer, Charles H. Laber, Doris Y. Melgarejo, Gina M. Chiarella, John P. Fackler Jr., Xinzhen Yang, Gary A. Baker, and Justin R. Walensky\*

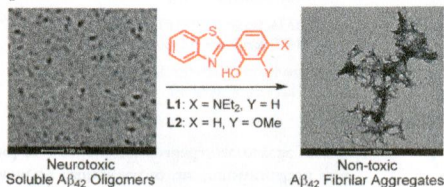
Novel dinuclear Cu(I) and Ag(I) formamidinate complexes have been synthesized, and their reactivity has been probed with carbon disulfide insertion producing tetranuclear or hexanuclear clusters. Only the 2,6-dimethylphenyl derivative of the dinuclear Cu(I) formamidinate complex is luminescent, unlike the Ag(I) and Au(I) analogs, and shows reversible quenching with  $\text{O}_2$ .



### Small Bifunctional Chelators That Do Not Disaggregate Amyloid $\beta$ Fibrils Exhibit Reduced Cellular Toxicity

Anuj K. Sharma, Jaekwang Kim, John T. Prior, Nicholas J. Hawco, Nigam P. Rath, Jungsu Kim, and Liviu M. Mirica\*

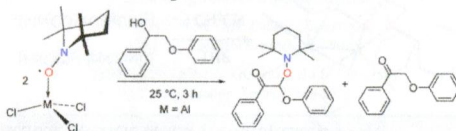
Two small bifunctional chelators, L1 and L2, were developed to specifically target the interactions of  $\text{Cu}^{2+}$  and  $\text{Zn}^{2+}$  ions with the  $\text{A}\beta_{42}$  peptide and to control the formation of neurotoxic  $\text{A}\beta_{42}$  aggregates. Unprecedentedly, these compounds promote the fibrillization of  $\text{Cu}^{2+}$ -stabilized soluble  $\text{A}\beta_{42}$  oligomers, and most importantly they drastically reduce the neurotoxicity of soluble  $\text{A}\beta_{42}$  species, in both the presence or absence  $\text{Cu}^{2+}$  ions



### Oxidation of Alcohols and Activated Alkanes with Lewis Acid-Activated TEMPO

Thuy-Ai D. Nguyen, Ashley M. Wright, Joshua S. Page, Guang Wu, and Trevor W. Hayton\*

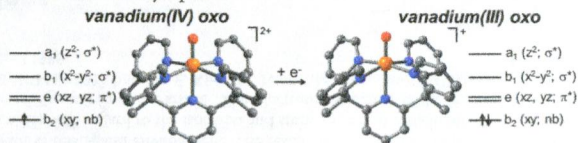
$\text{MCl}_3(\eta^1\text{-TEMPO})$  ( $\text{M} = \text{Fe}, \text{Al}$ , TEMPO = 2,2,6,6-tetramethylpiperidine-*N*-oxyl) can oxidize a variety of alcohols and activated alkanes, including 1-phenyl-2-phenoxyethanol. However, in some instances overoxidation of substrate is observed, concomitant with incorporation of TEMPO into the products of oxidation.



### A Well-Defined Terminal Vanadium(III) Oxo Complex

Amanda E. King, Michael Nippe, Mihail Atanasov,\* Teera Chantarojsiri, Curtis A. Wray, Eckhard Bill, Frank Neese,\* Jeffrey R. Long,\* and Christopher J. Chang\*

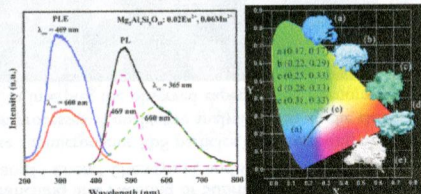
The synthesis, characterization, and computational analysis of a unique terminal mononuclear vanadium(III) oxo species is presented. This low-valent vanadyl, supported by the pentadentate polypyridyl ligand 2,6-bis[1,1-bis(2-pyridyl)ethyl]pyridine ( $\text{PYSMes}_2$ ), is reactive toward protons but not toward H- and O-atom-transfer reagents. Computations predict that further one-electron reduction of the vanadium(III) oxo species will result in ligand-based reduction, which has implications for the synthesis and reactivity of low-valent vanadyl species.



# Luminescence Properties and Energy Transfer of Eu/Mn-Coactivated $\text{Mg}_2\text{Al}_4\text{Si}_5\text{O}_{18}$ as a Potential Phosphor for White-Light LEDs

Jian Chen, Yangai Liu,\* Minghao Fang, and Zhaohui Huang

A series of color-tunable blue–white emitting  $\text{Mg}_2\text{Al}_4\text{Si}_5\text{O}_{18}$ :  $\text{Eu}^{2+}$ ,  $\text{Mn}^{2+}$  phosphors were prepared by a high-temperature solid-state reaction, and their luminescence properties were investigated. The energy-transfer mechanism was verified by the change in the luminescence spectra, decay curve, and energy-transfer efficiency of the phosphors, and the emission hue can be tuned from blue (0.17, 0.17) to bluish-green (0.22, 0.29) and finally to white (0.31, 0.33) emissions by properly varying the ratio of  $\text{Eu}^{2+}/\text{Mn}^{2+}$ .



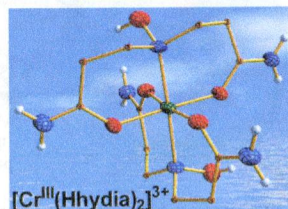
## 11404

DOI: 10.1021/ic501778d

# Interaction of Chromium(III) with a $N,N'$ -Disubstituted Hydroxylamine-(diamido) Ligand: A Combined Experimental and Theoretical Study

Petros A. Tziouris, Constantinos G. Tsiafoulis, Manolis Vlasios, Haralampos N. Miras,\* Michael P. Sigalas,\* Anastasios D. Keramidas,\* and Themistoklis A. Kabanos\*

Interaction of  $\text{trans}[\text{Cr}^{\text{III}}\text{Cl}_2(\text{H}_2\text{O})_4]\text{Cl} \cdot 2\text{H}_2\text{O}$  with the  $N,N'$ -disubstituted hydroxylamine-(diamido) ligand  $\text{H}(\text{hydia})$  gives  $[\text{Cr}^{\text{III}}(\text{H}(\text{hydia})_2)\text{Cl}_3]$  (**1**) as a major species. The X-ray structure analysis of **1** revealed that the chromium atom in  $[\text{Cr}^{\text{III}}(\text{H}(\text{hydia})_2)]^{3+}$  is bonded to two neutral tridentate  $\text{O}_2\text{N}_2\text{O}$ - $\text{H}(\text{hydia})$  ligands. The two intramolecular hydrogen bonds in  $[\text{Cr}(\text{H}(\text{hydia})_2)]^{3+}$  contribute to its stability.



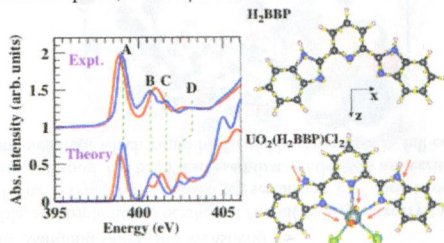
## 11415

DOI: 10.1021/ic501107a

# Bonding and Charge Transfer in Nitrogen-Donor Uranyl Complexes: Insights from NEXAFS Spectra

C. D. Pemmaraju,\* Roy Copping, Shuao Wang, Markus Janousch, Simon. J. Teat, Tolek Tylicszak, Andrew Canning, David K. Shuh, and David Prendergast

The electronic structures of recently synthesized uranyl ( $\{\text{UO}_2\}^{2+}$ ) complexes  $[\text{UO}_2(\text{H}_2\text{bbp})\text{Cl}_2]$ ,  $[\text{UO}_2(\text{Hbbp})(\text{Py})\text{Cl}]$ , and  $[\text{UO}_2(\text{bbp})(\text{Py})_2]$ , featuring the polarizable nitrogen-donor ligand 2,6-bis(2-benzimidazolyl)pyridine (bbp), are investigated through N K-edge NEXAFS spectroscopy measurements and first-principles simulations. Spectral features reveal both the polarization of the largely  $\pi$ -conjugated framework of the bbp ligand upon complexation with uranyl as well as trends in bbp–uranyl bonding across the three complexes, driven by chemical differences in the coordination shell of the uranyl moiety.

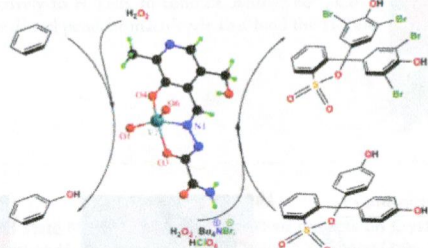




# Synthesis, Characterization, X-ray Crystal Structure, DFT Calculations, and Catalytic Properties of a Dioxidovanadium(V) Complex Derived from Oxamohydrazide and Pyridoxal: A Model Complex of Vanadate-Dependent Bromoperoxidase

Chandrima Das, Piyali Adak, Satyajit Mondal, Ryo Sekiya, Reiko Kuroda, Serge I. Gorelsky,\* and Shyamal Kumar Chattopadhyay\*

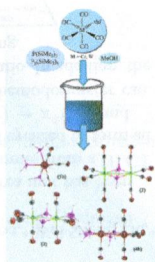
A vanadium(V) complex with the formula  $[\text{Et}_3\text{NH}][\text{V}^{\text{VO}_2}(\text{soxH-pydx})]$  with new tridentate ligand soxH-pydxH, obtained by condensation of oxamohydrazide and pyridoxal, is found to possess exceptionally high catalytic efficiency ( $k_{\text{cat}} = 26340 \text{ h}^{-1}$ ) in peroxidative bromination of the model substrate phenol red to bromophenol blue. The vanadium compound also efficiently catalyzes bromination of phenol and salicylaldehyde as well as oxidation of benzene to phenol by  $\text{H}_2\text{O}_2$ .



# Transition-Metal Complexes Containing Parent Phosphine or Phosphinyl Ligands and Their Use as Precursors for Phosphide Nanoparticles

Susanne Bauer, Cornelia Hunger, Michael Bodensteiner, Wilfried-Solo Ojo, Arnaud Cros-Gagneux, Bruno Chaudret, Céline Nayral, Fabien Delpéch, and Manfred Scheer\*

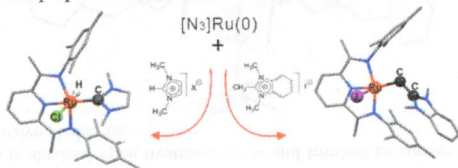
The P–H functional complexes  $[(\text{CO})_x\text{M}(\mu\text{-PH}_2)]_2$  ( $\text{M} = \text{Fe}$ ,  $x = 3$ ;  $\text{M} = \text{Cr}$ ,  $x = 4$ ) were synthesized in a one pot reaction by using  $\text{P}_2(\text{SiMe}_3)_4$ . Alternatively, photolysis of  $[(\text{CO})_3\text{M}(\text{PH}_3)_3]$  ( $\text{M} = \text{Cr}$  (**5a**),  $\text{M} = \text{W}$  (**5b**)) or of  $[(\text{CO})_x\text{M}(\mu\text{-PH}_2)]_2$  with  $\text{P}(\text{SiMe}_3)_3$  leads to  $[(\text{CO})_4\text{M}(\mu\text{-PH}_2)_2\text{M}(\text{CO})_3(\text{PH}_3)]$  ( $\text{M} = \text{Cr}$  (**3a**),  $\text{M} = \text{W}$  (**3b**)) and  $[(\text{CO})_4\text{W}(\mu\text{-PH}_2)_2\text{W}(\text{CO})_2(\text{PH}_3)_2]$  (**4b**). Complex **2a** was used as single source precursor for the stoichiometry-controlled synthesis of CrP nanoparticles.



# Synthesis of Tridentate 2,6-Bis(imino)pyridyl Ruthenium(II) Complexes with N-Heterocyclic Carbene Ligands: Activation of Imidazolium Salts

Hyojong Yoo\* and Donald H. Berry

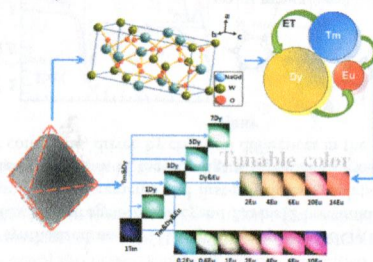
Tridentate 2,6-bis(imino)pyridyl ruthenium(II) complexes with hydride and N-heterocyclic carbene ligands were successfully synthesized through the activation of imidazolium salts by low-valent Ru(0) complexes. A zwitterionic ruthenium complex with an ylidic ligand, bis(imino)pyridyl ruthenium-(2-methyleneimidazoline) complex, was also isolated. Formation mechanisms of these complexes are proposed and discussed.



# Single-Component and Warm-White-Emitting Phosphor $\text{NaGd}(\text{WO}_4)_2\text{Tm}^{3+}$ , $\text{Dy}^{3+}$ , $\text{Eu}^{3+}$ : Synthesis, Luminescence, Energy Transfer, and Tunable Color

Yan Liu, Guixia Liu,\* Jinxian Wang, Xiangting Dong, and Wensheng Yu

$\text{Tm}^{3+}$ ,  $\text{Dy}^{3+}$ , and  $\text{Eu}^{3+}$  single, double, and triple doped octahedral  $\text{NaGd}(\text{WO}_4)_2$  microcrystals were prepared by hydrothermal process. The energy migration of  $\text{Tm}^{3+}$ – $\text{Dy}^{3+}$ – $\text{Eu}^{3+}$ , utilized for sensitizing  $\text{Eu}^{3+}$  ions besides compensating the red component at low  $\text{Eu}^{3+}$  doping concentration, has been discussed first. Under 365 nm excitation, the color-tunable emissions are realized, especially warm white emission, which could be favorable candidates in full-color phosphors for nUV-LEDs.



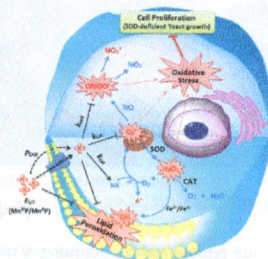
## 11467

DOI: 10.1021/ic501329p

# Rational Design of Superoxide Dismutase (SOD) Mimics: The Evaluation of the Therapeutic Potential of New Cationic Mn Porphyrins with Linear and Cyclic Substituents

Artak Tovmasyan, Sebastian Carballal, Robert Ghazaryan, Lida Melikyan, Tin Weitner, Clarissa G. C. Maia, Julio S. Reboucas, Rafael Radi, Ivan Spasojevic, Ludmil Benov, and Ines Batinic-Haberle\*

Mn porphyrins (MnPs) with linear and cyclic substituents were synthesized and evaluated for their *in vitro* and *in vivo* therapeutic potential. The  $E_{1/2}$  of the  $\text{Mn}^{\text{III}}\text{P}/\text{Mn}^{\text{II}}\text{P}$ , which parallels  $k_{\text{cat}}(\text{O}_2^{\bullet-})$  and  $k_{\text{red}}(\text{ONOO}^-)$ , controls the magnitude of lipid peroxidation inhibition by MnPs, and their ability to protect SOD-deficient yeast growth and to catalyze ascorbate oxidation.



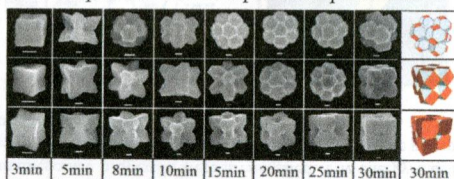
## 11484

DOI: 10.1021/ic501368y

# Local Supersaturation Dictated Branching and Faceting of Submicrometer PbS Particles with Cubic Growth Habit

Mingzhu Liu, Mei Leng, Dan Liu, Fanglin Chen,\* Chengyu Li, and Cheng Wang\*

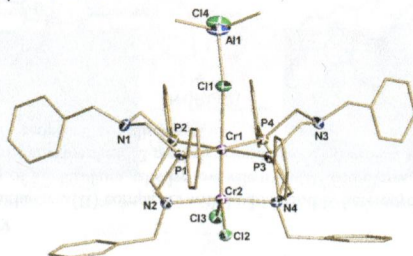
Three kinds of hierarchical cubelike submicrometer PbS particles that consisting of truncated octahedrons, cuboctahedrons, and cubes were obtained in solution processes. A qualitative analysis based on the classical nucleation theory coupled with the crystal growth theory is employed to interpret the observed experimental phenomena.



### Chromium–Chromium Interaction in a Binuclear Mixed-Valent Cr<sup>I</sup>–Cr<sup>II</sup> Complex

Ahmed Alzamly, Serge I. Gorelsky, Sandro Gambarotta,\* Ilia Korobkov, Jennifer Le Roy, and Muralee Murugesu

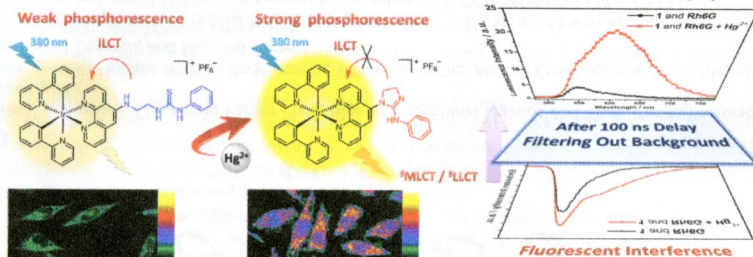
A mixed-valent Cr<sup>I</sup>–Cr<sup>II</sup> species is obtained upon treatment of di- and trivalent precursors with Me<sub>3</sub>Al. The presence of a Cr–Cr double bond prevents catalytic behavior.



### Turn-on Phosphorescent Chemodosimeter for Hg<sup>2+</sup> Based on a Cyclometalated Ir(III) Complex and Its Application in Time-Resolved Luminescence Assays and Live Cell Imaging

Jia-Xi Ru, Li-Ping Guan, Xiao-Liang Tang,\* Wei Dou, Xiang Yao, Wan-Min Chen, Ya-Ming Liu, Guo-Lin Zhang,\* Wei-Sheng Liu,\* Yue Meng, and Chun-Ming Wang

A novel phosphorescent chemodosimeter based on a cyclometalated Ir(III) complex containing a thiourea unit has been successfully synthesized and characterized. The complex exhibits weak phosphorescence, which mainly comes from a <sup>3</sup>ILCT excited state according to TD-DFT calculations. In the presence of Hg<sup>2+</sup>, the thiourea unit of the ligand is cyclized to form an imidazoline moiety, resulting in stronger phosphorescence attributed to an admixture of <sup>3</sup>MLCT (dπ(Ir) → π\*(L)) and <sup>3</sup>LLCT (π(ppy) → π\*(L)). Furthermore, by using a time-resolved photoluminescence technique, the chemodosimeter can eliminate the interference from the short-lived background fluorescence and improve the signal-to-noise ratio. In addition, the chemodosimeter can be used to monitor Hg<sup>2+</sup> effectively in living cells by confocal luminescence imaging.

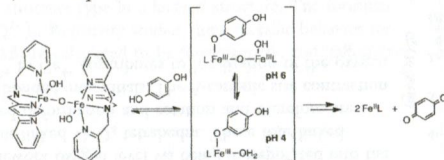




### Speciation of Ferric Phenoxide Intermediates during the Reduction of Iron(III)- $\mu$ -Oxo Dimers by Hydroquinone

William D. Kerber,\* Kaitlyn A. Perez, Chuqiao Ren, and Maxime A. Siegler

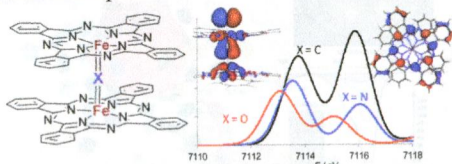
Iron(III) is chelated by tris(pyridylmethyl)amine to form  $[\text{Fe}(\text{TPA})]^{3+}(\text{aq})$  at low pH and the  $\mu$ -oxo dimer  $[\text{Fe}_2(\mu\text{-O})(\text{TPA})_2(\text{H}_2\text{O})_2]^{+4}$  above pH 3. The dimer is a diprotic acid with  $\text{p}K_{\text{a}1,2} = 4.38$  and 5.26. Addition of a phenol displaces solvent from the labile coordination sites to form ferric phenoxide complexes that have a pH-dependent speciation. When the incoming phenol is hydroquinone, the complex is reduced to iron(II) with concomitant formation of benzoquinone.



### X-ray Absorption and Emission Spectroscopies of X-Bridged Diiron Phthalocyanine Complexes $(\text{FePc})_2\text{X}$ (X = C, N, O) Combined with DFT Study of $(\text{FePc})_2\text{X}$ and Their High-Valent Diiron Oxo Complexes

Cedric Colombar, Evgenij V. Kudrik, Valerie Briois, Janine C. Shwarbrick, Alexander B. Sorokin,\* and Pavel Afanasiev\*

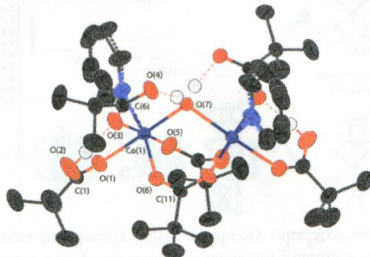
A series of  $\mu$ -bridged diiron phthalocyanine complexes  $[\text{PcFeXFePc}]$  (X = C, N, O) was studied by core hole spectroscopies at the Fe K edge, including EXAFS, XES, and resonant XANES. Spectroscopy-calibrated DFT calculations provide a coherent description of the molecular geometry and electronic structure. Theory calculations extrapolated to the high-valent putative oxo complexes of three diiron species allow explaining of the outstanding catalytic activity of the  $\mu$ -nitrido complex as compared to its  $\mu$ -oxo and  $\mu$ -carbido counterparts.



### Relationships between Electron Density and Magnetic Properties in Water-Bridged Dimetal Complexes

Jacob Overgaard,\* James P. S. Walsh, Venkatesha R. Hathwar, Mads R. V. Jørgensen, Christina Hoffman, Jamie A. Platts, Ross Piltz, and Richard E. P. Winpenny

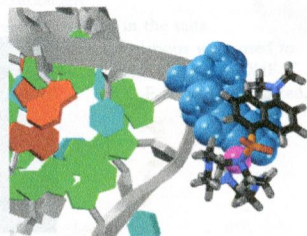
The experimental and theoretical electron density distributions in two structurally similar transition metal dimers (M = Ni, Co; see Figure) were analyzed using the atoms-in-molecules (AIM) approach, and selected properties related to the chemical bonding are compared to measured intramolecular magnetic exchange interaction parameters.



### Selective Binding of $\text{Zn}^{2+}$ Complexes to Human Telomeric G-Quadruplex DNA

Kevin E. Sifers, Matthew A. Fountain,\* and Janet R. Morrow\*

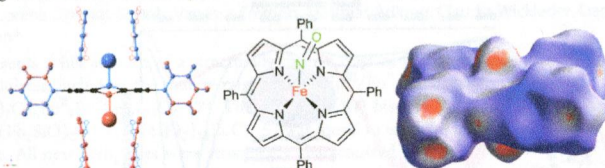
Two  $\text{Zn}^{2+}$  macrocyclic complexes are studied as binders for thymine-containing sequences in the loops of the human telomeric (H-Telo) G-quadruplex. Data support the selective binding of a  $\text{Zn}^{2+}$  complex with a dansyl pendent to thymine nucleobases in the lateral loops of H-Telo. The  $\text{Zn}^{2+}$  complex containing an acridine pendent binds tightly but nonselectively to H-Telo. In contrast, neither  $\text{Fe}^{2+}$ ,  $\text{Co}^{2+}$ , nor  $\text{Cu}^{2+}$  forms complexes with the dansyl pendent macrocycle that bind the H-Telo G-quadruplex.



### Intermolecular Interactions in Solid-State Metalloporphyrins and Their Impacts on Crystal and Molecular Structures

Seth C. Hunter, Brenda A. Smith, Christina M. Hoffmann,\* Xiaoping Wang,\* Yu-Sheng Chen, Garry J. McIntyre, and Zi-Ling Xue\*

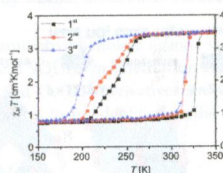
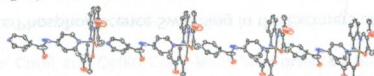
Hirshfeld surface analyses reveal *intermolecular* interactions in the crystals of  $[\text{M}(\text{TPP})\text{X}]$  ( $\text{TPP}^{2-}$  = *meso*-tetraphenylporphyrinate;  $\text{M} = \text{Fe}$ ,  $\text{X} = \text{Cl}$ ;  $\text{M} = \text{Co}$ ,  $\text{Fe}$ ,  $\text{X} = \text{NO}$ ) lead to phase changes in  $[\text{M}(\text{TPP})(\text{NO})]$  and a gradual structural change in  $[\text{Fe}(\text{TPP})\text{Cl}]$ , allowing reversible thermal compression at lower temperatures.



### Large Thermal Hysteresis for Iron(II) Spin Crossover Complexes with *N*-(Pyrid-4-yl)isonicotinamide

Charles Lochenie, Wolfgang Bauer, Antoine P. Railliet, Stephan Schlamp, Yann Garcia,\* and Birgit Weber\*

Iron(II) spin crossover materials with wide hysteresis loops are obtained due to a combination of short-range (hydrogen bonds) and long-range (coordination polymer) interactions.

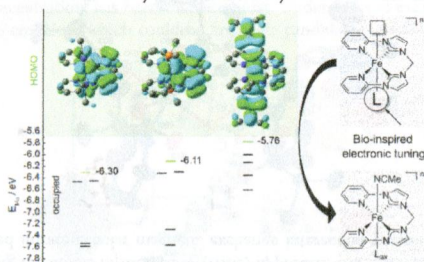




# Making Oxidation Potentials Predictable: Coordination of Additives Applied to the Electronic Fine Tuning of an Iron(II) Complex

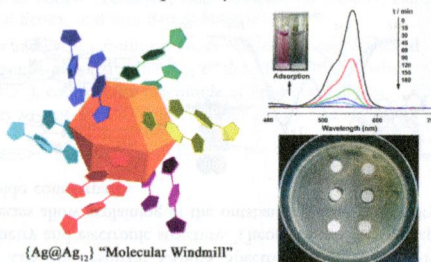
Stefan Haslinger, Jens W. Kück, Eva M. Hahn, Mirza Cokoja, Alexander Pöthig, Jean-Marie Basset, and Fritz E. Kühn\*

This work examines the impact of axially coordinating additives on the electronic structure of an iron(II) complex, inspired by apical coordination in biological systems. The predictability of oxidation potentials is demonstrated, based on a correlation between cyclic voltammetry experiments and density functional theory calculated molecular orbital energies.



# A Nanosized $\{Ag@Ag_{12}\}$ "Molecular Windmill" Templated by Polyoxometalates Anions

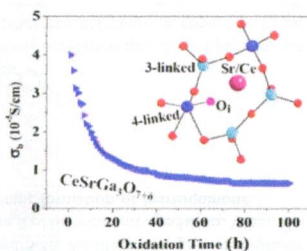
Lei Wang, Weiting Yang, Wei Zhu, Xingang Guan, Zhigang Xie,\* and Zhong-Ming Sun\*  
A nanosized molecular windmill-shaped polynuclear Ag cluster with intriguing  $\{M@M_{12}\}$ -type cuboctahedral topology with effective photocatalytic and antibacterial activities templated by POMs.



# Localization of Oxygen Interstitials in $CeSrGa_3O_{7+\delta}$ Melilite

Jungu Xu, Xiaojun Kuang,\* Emmanuel Véron, Mathieu Allix, Matthew R. Suchomel, Florence Porcher, Chaolun Liang, Fengjuan Pan, and Mingmei Wu\*

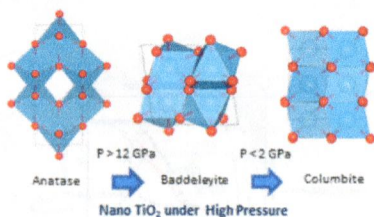
The excess oxygen atoms from the oxidation of  $Ce^{3+}$  to  $Ce^{4+}$  in  $CeSrGa_3O_{7+\delta}$  melilite are localized at the framework oxygen level via being incorporated into the coordination environment of four-linked  $GaO_4$  tetrahedra. These four-linked tetrahedral units have constrained deformation and rotation and therefore are unfavorable for transporting the bonded interstitials. The A-cationic size contraction induced by the oxidation of  $Ce^{3+}$  to  $Ce^{4+}$  contributes to localization of the oxygen interstitials.



### Compressibility and Structural Stability of Nanocrystalline $\text{TiO}_2$ Anatase Synthesized from Freeze-Dried Precursors

Catalin Popescu,\* Juan Angel Sans, Daniel Errandonea, Alfredo Segura, Regina Villanueva, and Fernando Sapiña

The HP structural behavior of anatase-type nanoparticles was studied using X-ray diffraction under hydrostatic and quasi-hydrostatic conditions up to 25 GPa. Experiments were carried out by the first time under hydrostatic conditions. An anatase-baddeleyite phase transition is found under compression. On decompression a columbite-type structure is recovered as a metastable phase. Possible mechanisms for the observed phase transition are discussed. Results are compared with previous HP studies on anatase nanoparticles.



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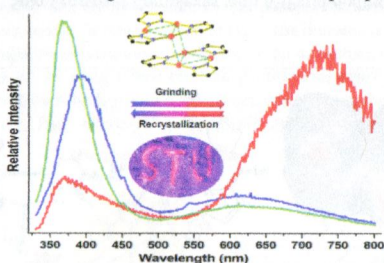


DOI: 10.1021/ic5016687

### Mechanically Triggered Fluorescence/Phosphorescence Switching in the Excimers of Planar Trinuclear Copper(I) Pyrazolate Complexes

Qiong Xiao, Ji Zheng, Mian Li, Shun-Ze Zhan, Jun-Hao Wang, and Dan Li\*

A series of cyclic trinuclear copper(I) pyrazolate clusters with auxochrome-appended ligands have been found to exhibit unusual luminescence mechanochromism, featuring switchable fluorescent and phosphorescent bands in dual emission.



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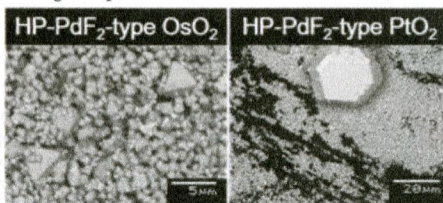


DOI: 10.1021/ic501770g

### Synthesis, Crystal Structure, and Electronic Properties of High-Pressure $\text{PdF}_2$ -Type Oxides $\text{MO}_2$ ( $M = \text{Ru}, \text{Rh}, \text{Os}, \text{Ir}, \text{Pt}$ )

Yuichi Shirako,\* Xia Wang, Yoshihiro Tsujimoto, Kie Tanaka, Yanfeng Guo, Yoshitaka Matsushita, Yoshihiro Nemoto, Yoshio Katsuya, Youguo Shi, Daisuke Mori, Hiroshi Kojitani, Kazunari Yamaura, Yoshiyuki Inaguma, and Masaki Akaogi

The polycrystalline  $\text{MO}_2$ 's (HP- $\text{PdF}_2$ -type  $\text{MO}_2$ ,  $M = \text{Rh}, \text{Os}, \text{Pt}$ ) with high-pressure  $\text{PdF}_2$  compounds were successfully synthesized under high-pressure and high-temperature conditions for the first time, to the best of our knowledge.

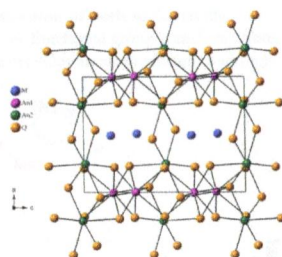




# Syntheses, Crystal Structures, Transport Properties, and Theoretical Studies of Five Members of the $\text{MAN}_2\text{Q}_5$ Family: $\text{SrU}_2\text{S}_5$ , $\text{BaU}_2\text{Se}_5$ , $\text{PbU}_2\text{S}_5$ , $\text{BaTh}_2\text{S}_5$ , and $\text{BaU}_2\text{Te}_5$

Jai Prakash, Mariya S. Tarasenko, Adel Mesbah, Sébastien Lebègue, Christos D. Malliakas, and James A. Ibers\*

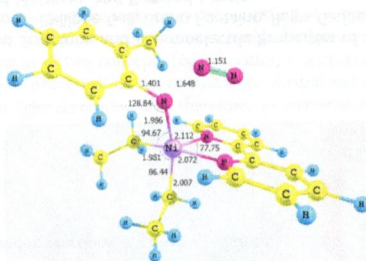
$\text{SrU}_2\text{S}_5$ ,  $\text{BaU}_2\text{Se}_5$ ,  $\text{PbU}_2\text{S}_5$ ,  $\text{BaTh}_2\text{S}_5$ , and  $\text{BaU}_2\text{Te}_5$  of the  $\text{MAN}_2\text{Q}_5$  family were obtained by solid-state reactions. The sulfides and selenides are isostructural and crystallize in the  $\text{PbU}_2\text{Se}_5$  structure type in a three-dimensional structure; in contrast,  $\text{BaU}_2\text{Te}_5$  crystallizes in the  $(\text{NH}_4)\text{Pb}_2\text{Br}_5$  structure type in a layered structure. The formulas charge balance as  $\text{M}^{2+}(\text{An}^{4+})_2(\text{Q}^{2-})_5$ . Resistivity studies show metallic behavior for  $\text{SrU}_2\text{S}_5$ ,  $\text{BaU}_2\text{Se}_5$ , and  $\text{PbU}_2\text{S}_5$ . All are calculated to be ferromagnetic and half-metals.



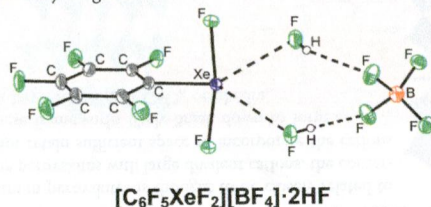
# The Curious Case of Mesityl Azide and Its Reactivity with $\text{bpyNiEt}_2$

Brooke M. Otten, Travis M. Figg, and Thomas R. Cundari\*

DFT analysis of the reaction of  $\text{bpyNiEt}_2$  with  $\text{ArN}_3$  is performed for *ortho*- and *para*-tolyl, *meta*-xylyl, and mesityl azide ( $\text{MesN}_3$ ). Of interest are the different products obtained for the latter (ethylene, butane, azomesitylene, *N*-mesityl-ethylamine) versus the other reagents:  $(\text{bpyNi}(\text{N}(\text{Ar})\text{Et})(\text{Et}))$ . Metal-free reactions do not differentiate  $\text{MesN}_3$  from other azides. Calculated Ni–N/C bond energies suggest the role of radicals in discriminating among these reactions, and that their concentration will be greatest for  $\text{MesN}_3$ .



The X-ray crystal structures of the first Xe(IV)–C bonded species,  $[\text{C}_6\text{F}_5\text{XeF}_2]^+$ , were determined in the salts  $[\text{C}_6\text{F}_5\text{XeF}_2][\text{BF}_4]$ ,  $[\text{C}_6\text{F}_5\text{XeF}_2][\text{BF}_4] \cdot 2\text{HF}$ , and  $[\text{C}_6\text{F}_5\text{XeF}_2][\text{BF}_4] \cdot 1.5\text{NCCCH}_3$ . Quantum-chemical calculations were used to arrive at detailed vibrational (Raman) assignments, as well as the NBO analyses and MO descriptions of the  $[\text{C}_6\text{F}_5\text{XeF}_2]^+$  cation and the isoelectronic  $\text{C}_6\text{F}_5\text{IF}_2$  molecule. The  $^{19}\text{F}$  and  $^{129}\text{Xe}$  NMR spectra of  $[\text{C}_6\text{F}_5\text{XeF}_2]^+$  and  $\text{C}_6\text{F}_5\text{IF}_2$  were simulated, and their spin–spin couplings were fully assigned.



Janek Rausch, Volker Lorenz, Cristian G. Hrib, Vanessa Frettlöh, Matthias Adlung, Claudia Wickleder, Liane Hilfert, Peter G. Jones, and Frank T. Edelmann\*

Carmen Martín, Christopher J. Whiteoak, Eddy Martín, Eduardo C. Escudero-Adán, José Ramón Galán-Mascarós, and Arjan W. Kleij<sup>\*</sup>

**Co(II)**

**Co(III)**

DMPA or  
Bipyridine

7. R = Me, L = (DMPA)<sub>2</sub>  
8. R = tBu, L = DMAP  
9. R = Me, L = 2,2'-bipy

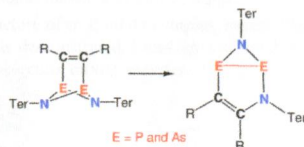
4. R = H  
5. R = Me



# A New Class of Azadipnictiridines Generated by an Unusual Rearrangement Reaction

Alexander Hinz, Axel Schulz,\* Wolfram W. Seidel, and Alexander Villinger

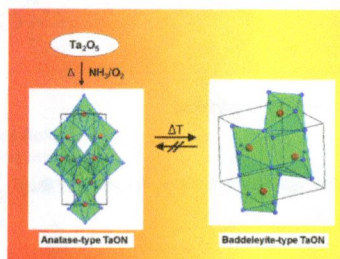
Dipnictadiazanediyls add readily alkynes affording [2.1.1]bicycles, which rearrange slowly to [3.1.0]bicycles resulting in the formation of novel azadipnictiridines.



# Synthesis and Crystal Structure of $\delta$ -TaON, a Metastable Polymorph of Tantalum Oxide Nitride

Tobias Lütke, Alexander Schmidt, Caren Göbel, Anna Fischer, Nils Becker, Christoph Reimann, Thomas Bredow, Richard Dronskowski, and Martin Lerch\*

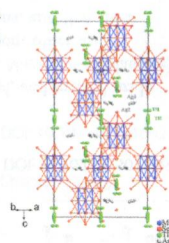
Tetragonal anatase-type  $\delta$ -TaON was prepared by reaction of gaseous ammonia with an amorphous tantalum oxide precursor. At temperatures between 800 and 850 °C an irreversible phase transformation to baddeleyite-type  $\beta$ -TaON is observed. Quantum-chemical calculations confirm the dynamic stability of  $\delta$ -TaON. It was found that an N/O distribution with maximal N–N distances is most stable. The calculated band edge energies indicate that  $\delta$ -TaON is a promising photocatalytic material for redox reactions, e.g., water splitting.



# X-ray Characterization, Electronic Band Structure, and Thermoelectric Properties of the Cluster Compound $Ag_2Tl_2Mo_9Se_{11}$

Rabih Al Rahal Al Orabi, Patrick Gougeon,\* Philippe Gall, Bruno Fontaine, Régis Gautier, Malika Colin, Christophe Candolfi, Anne Dauscher, Jiri Hejtmanek, Bernard Malaman, and Bertrand Lenoir

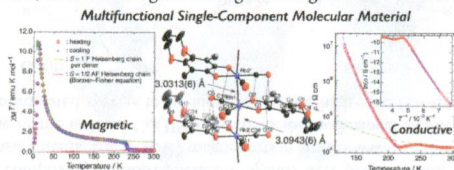
The novel Mo-based cluster compound  $Ag_2Tl_2Mo_9Se_{11}$  shows glass-like thermal conductivity due to a strong disorder induced by the Ag and Tl atoms, leading to interesting thermoelectric properties at high temperatures.



# Multifunctional One-Dimensional Rhodium(II)–Semiquinonato Complex: Substituent Effects on Crystal Structures and Solid-State Properties

Minoru Mitsumi,\* Shoji Ohtake, Yuki Kakuno, Yuuki Komatsu, Yoshiaki Ozawa, Koshiro Toriumi, Nobuhiro Yasuda, Nobuaki Azuma, and Yuji Miyazaki

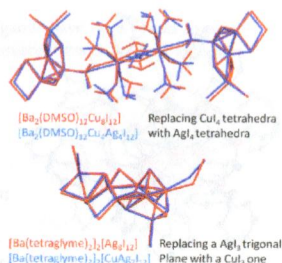
The crystal structures and magnetic and conductive properties of two new one-dimensional (1D) rhodium(II)–semiquinonato complexes **4** and **5** have been investigated. Compound **4** exhibits intriguing changes in its magnetic and conducting properties, and these changes are accompanied by a first-order phase transition. In comparison, compound **5**, which assumes a strongly dimerized structure in the 1D chain, exhibits no sign of strong ferromagnetic interactions and is an insulator.



# Novel Barium–Organic Incorporated Iodometalates: Do They Have Template Properties for Constructing Rare Heterotrimetallic Hybrids?

Shashank Mishra,\* Erwann Jeanneau, Gilles Ledoux, and Stéphane Daniele\*

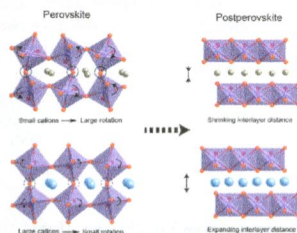
Novel barium-organic incorporated iodocuprates and -argentates retain their structural motifs when a  $MI_4$  tetrahedral/ $MI_3$  trigonal-coplanar unit is replaced by another  $M'I_4$ / $M'I_3$  unit. Such structure-directing properties are, however, not observed when a  $MI_4$ / $MI_3$  unit is replaced by a  $M'I_6$  octahedron, which requires molecular rearrangement.



# Postperovskite Phase Transition of $ZnGeO_3$ : Comparative Crystal Chemistry of Postperovskite Phase Transition from Germanate Perovskites

Hitoshi Yusa,\* Taku Tsuchiya, Masaki Akaogi, Hiroshi Kojitani, Daisuke Yamazaki, Naohisa Hirao, Yasuo Ohishi, and Takumi Kikigawa

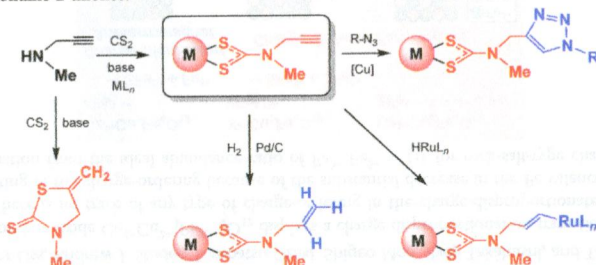
The rotation of the  $GeO_6$  octahedra in perovskites is thought to be closely related to the postperovskite transition. In the perovskites with large divalent cations, the corner-sharing  $GeO_6$  frameworks could not retain sufficient space to incorporate the cations by increasing the tilting angle. These frameworks likely break down to form a postperovskite structure without a large rotation of  $GeO_6$  octahedra.



# Multimetallc Complexes and Functionalized Nanoparticles Based on Unsymmetrical Dithiocarbamate Ligands with Allyl and Propargyl Functionality

Venesia L. Hurtubise, James M. McArdle, Saira Naeem, Anita Toscani, Andrew J. P. White, Nicholas J. Long,\* and James D. E. T. Wilton-Ely\*

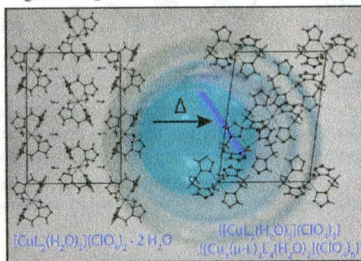
Unsymmetrical dithiocarbamate ligands prove to be versatile starting points for the preparation of both multimetallic complexes and surface-functionalized gold nanoparticles. The ligands provide a platform for functional group transformations of the pendant alkene or alkyne functionality, while cyclization of the propargyl derivative provides an atom-efficient and high-yielding route to thiazolidine-2-thiones.



# Cocrystallization of Photosensitive Energetic Copper(II) Perchlorate Complexes with the Nitrogen-rich Ligand 1,2-Di(1H-tetrazol-5-yl)ethane

Jürgen Evers, Ivan Gospodinov, Manuel Joas, Thomas M. Klapötke,\* and Jörg Stierstorfer

The preparation and characterization of an energetic ionic cocrystal consisting of copper(II) as a central metal ion, 1,2-di(1H-tetrazol-5-yl)ethane as a ligand, and perchlorate as a counterion is, beside further coordination compounds, presented. This ionic cocrystal represents the combination of two concepts (complexation and cocrystallization) in designing energetic materials. Further, different coordination spheres within the cocrystal (octahedral and square pyramidal) were shown by UV/vis/NIR spectroscopy to result in high light absorption and, therefore, an increased sensitivity toward laser irradiation.

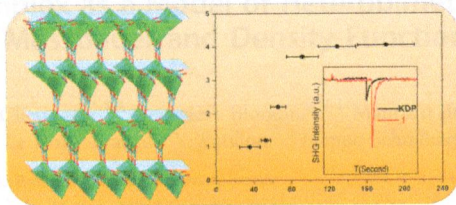




### An Acentric Calcium Borate $\text{Ca}_2[\text{B}_5\text{O}_9](\text{OH})\cdot\text{H}_2\text{O}$ : Synthesis, Structure, and Nonlinear Optical Property

Qi Wei, Jian-Wen Cheng,\* Chao He, and Guo-Yu Yang\*

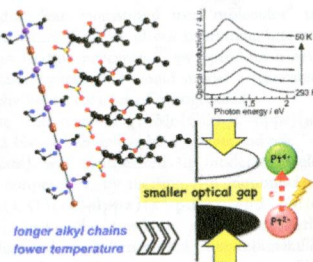
A new acentric hilgardite-type calcium borate,  $\text{Ca}_2[\text{B}_5\text{O}_9](\text{OH})\cdot\text{H}_2\text{O}$  (**1**), was made under solvothermal condition. Its framework features a denser net with interpenetrated *pcu* B–O net and *dia* Ca–O net. **1** is a potential deep-UV NLO material with a strong SHG response  $\sim 3$  times that of  $\text{KH}_2\text{PO}_4$  (KDP) and a short UV cutoff edge below 200 nm.



### Continuous Control of Optical Gaps in Quasi-One-Dimensional Bromide-Bridged Platinum Complexes by Utilizing Chemical Pressure

Shohei Kumagai, Hiroaki Iguchi, Shinya Takaishi,\* Brian K. Breedlove, Masahiro Yamashita,\* Hiroyuki Matsuzaki, Hiroshi Okamoto, Kenichi Kato, and Masaki Takata

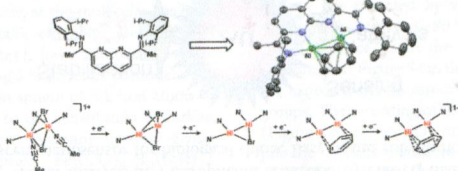
Bromide-bridged Pt chain complexes,  $[\text{Pt}(\text{en})_2\text{Br}](\text{C}_n\text{-Y})_2\cdot\text{H}_2\text{O}$  ( $\text{en}$  = ethylenediamine;  $\text{C}_n\text{-Y}$  = dialkyl sulfosuccinate;  $n$  = the number of carbon atoms), were synthesized. The Pt–Pt distances and the optical gaps depended on  $n$  and  $T$ . In addition,  $[\text{Pt}(\text{en})_2\text{Br}](\text{C}_{12}\text{-Y})_2\cdot\text{H}_2\text{O}$  had the shortest Pt–Pt distance among the bromide-bridged Pt chain complexes reported so far.



### Dinuclear Nickel Complexes in Five States of Oxidation Using a Redox-Active Ligand

You-Yun Zhou, Douglas R. Hartline, Talia J. Steiman, Phillip E. Fanwick, and Christopher Uyeda\*

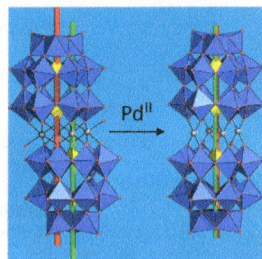
Dinuclear Ni complexes were synthesized using a redox-active naphthyridine–diimine (NDI) ligand system. Cyclic voltammetry experiments reveal rich, reversible redox chemistry for this platform, enabling access to a series of five (NDI) $\text{Ni}_2$  complexes related by electron transfer. The members of this series were characterized using structural and spectroscopic methods in combination with DFT models. The interplay between metal- and ligand-centered redox activity is discussed.



# Tetrapalladium-Containing Polyoxotungstate [Pd<sup>II</sup><sub>4</sub>( $\alpha$ -P<sub>2</sub>W<sub>15</sub>O<sub>56</sub>)<sub>2</sub>]<sup>16-</sup>: A Comparative Study

Natalya V. Izarova, Raisa I. Maksimovskaya, Sabine Willbold, and Paul Kögerler\*

Pd<sup>II</sup>-based sandwich complexes of lacunary {P<sub>2</sub>W<sub>15</sub>}<sub>2</sub>-type polyoxotungstates differ markedly from classical dimeric {M<sub>4</sub>(P<sub>2</sub>W<sub>15</sub>)<sub>2</sub>} structures incorporating octahedral first-row transition metal M<sup>II</sup> centers, form syn and anti isomers in a characteristic 2:1 ratio, and exhibit high stability in solution.



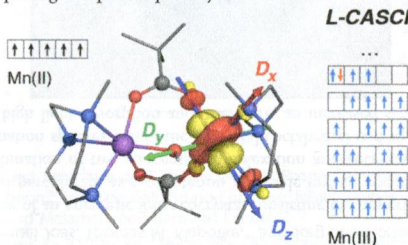
## 11785

DOI: 10.1021/ic502081c

# A First-Principles Approach to the Calculation of the on-Site Zero-Field Splitting in Polynuclear Transition Metal Complexes

Marius Retegan, Nicholas Cox, Dimitrios A. Pantazis, and Frank Neese\*

A new high-level multireference method is developed for the calculation of the on-site zero-field splitting parameters. The applicability of the method is demonstrated using a series of Mn(III)-containing complexes of increasing nuclearity, including the inorganic cluster of the water-splitting complex of photosystem II.



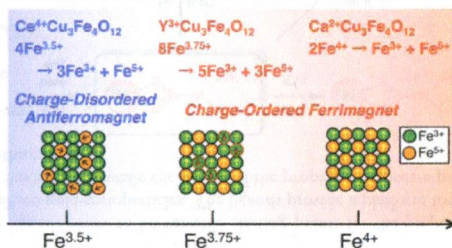
## 11794 **S**

DOI: 10.1021/ic502138v

# Charge-Order Melting in Charge-Disproportionated Perovskite CeCu<sub>3</sub>Fe<sub>4</sub>O<sub>12</sub>

Ikuya Yamada,\* Hidenobu Etani, Makoto Murakami, Naoaki Hayashi, Takateru Kawakami, Masaichiro Mizumaki, Shigenori Ueda, Hideki Abe, Klaus-Dieter Liss, Andrew J. Studer, Tomoatsu Ozaki, Shigeo Mori, Ryoji Takahashi, and Tetsuo Irfune

A novel quadruple perovskite oxide Ce<sup>4+</sup>Cu<sup>2+</sup><sub>3</sub>Fe<sup>3.5+</sup><sub>4</sub>O<sub>12</sub> displays a charge disproportionation transition of 4Fe<sup>3.5+</sup> → 3Fe<sup>3+</sup> + Fe<sup>5+</sup> below ~270 K. There is no trace of any type of charge-ordering in the charge-disproportionated CeCu<sub>3</sub>Fe<sub>4</sub>O<sub>12</sub> phase, demonstrating the melting of the charge-ordering because of the substantial decrease in the Fe valence from tetravalency and the resulting large deviation from the ideal abundance ratio of Fe<sup>3+</sup>:Fe<sup>5+</sup> = 1:1 for rock-salt-type charge-ordering.



**Stabilization of AuNPs by Monofunctional Triazole Linked to Ferrocene, Ferricenium, or Coumarin and Applications to Synthesis, Sensing, and Catalysis**

Na Li, Pengxiang Zhao, Maria E. Igartua, Amalia Rapakousiou, Lionel Salmon, Sergio Moya, Jaime Ruiz, and Didier Astruc\*

Ferrocene and coumarin triazole (Cou-trz) were utilized to stabilize gold nanoparticles (AuNPs). The first ferricenium-stabilized AuNP was directly obtained by the reaction between the trz-linked ferrocene and  $\text{HAuCl}_4$ . Ferrocene- and ferricenium-appended **AuNP-1** and **AuNP-2** reversibly interconvert. The trz-linked ferricenium chloride stabilization of **AuNP-2** is also characterized by slower catalysis in 4-nitrophenol reduction by  $\text{NaBH}_4$  than with other trz-AuNPs. **AuNP-3** capped with Cou-trz is a fluorescent nanosensor for biological thiols. Easy ligand substitution was also applied to dendrimer synthesis.

