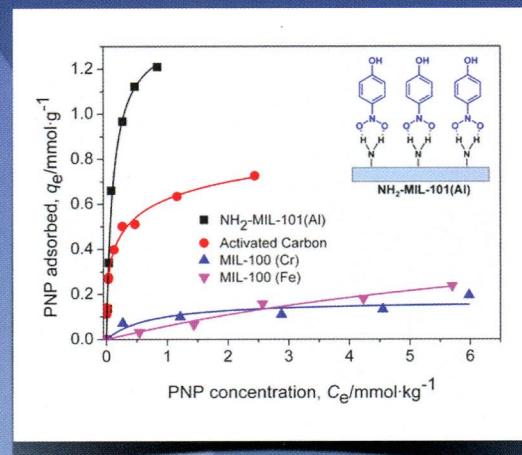
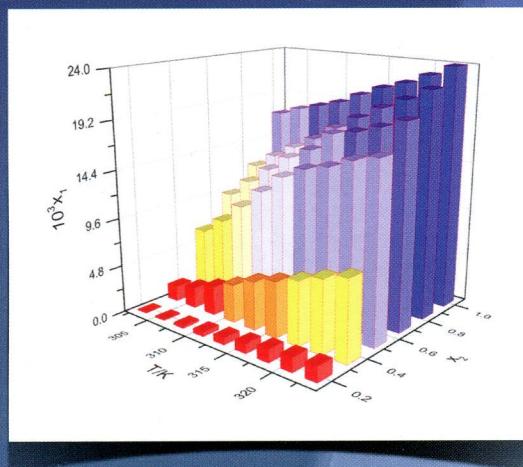
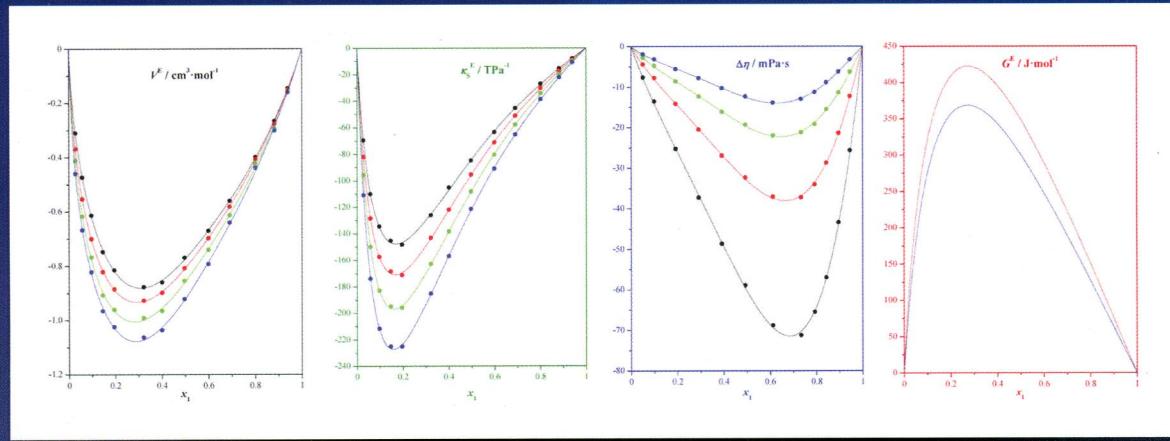




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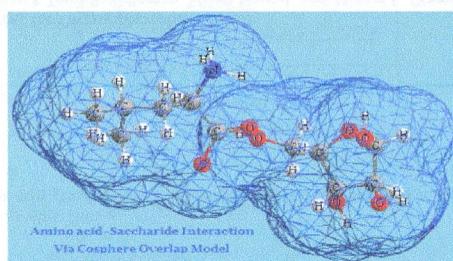
ON THE COVER: The images shown on the cover were taken from papers in this issue: (top) Graphs representing the thermophysical characterization of the binary mixture 1-propylpyridinium tetrafluoroborate with methanol (see DOI: 10.1021/je401120a). (middle left) Solubilities of ADCP in an ethanol + water mixture determined from (303.15 to 323.15) K by the synthetic method with a laser monitoring observation technique (see DOI: 10.1021/je401053y). Adsorption equilibria for phenol and PNP from water of three metal-organic frameworks. The graph shows the exceptional adsorption capacity of NH₂-MIL-101(Al) for PNP (middle right) (see DOI: 10.1021/je4010239). Optical high-pressure stainless steel equilibrium cell used for the determination of vapor-liquid equilibria for the binary mixture of ester diethyl fumarate with CO₂. (bottom) (see DOI: 10.1021/je4010849).

Articles

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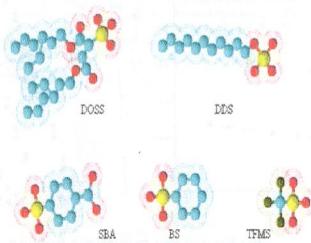
Partial Molar Volumes and Isentropic Compressibilities of Some Saccharides in Aqueous Solutions of Leucine at Different Temperatures

Suvarcha Chauhan* and Kuldeep Kumar

dx.doi.org/10.1021/je400977r1385 

Effect of Temperature and Anion on Densities, Viscosities, and Refractive Indices of 1-Octyl-3-propanenitrile Imidazolium-Based Ionic Liquids

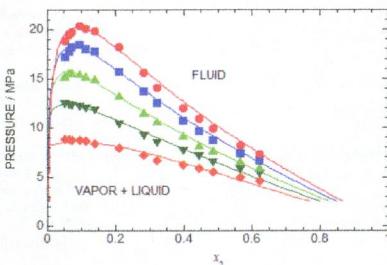
Abobakr K. Ziyada* and Cecilia D. Wilfred

dx.doi.org/10.1021/je400824z

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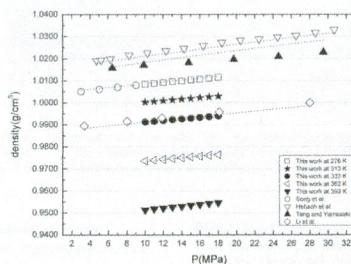
Cloud-Point and Bubble-Point Measurement for the Poly(2-butoxyethyl acrylate) + Cosolvent Mixture and 2-Butoxyethyl Acrylate in Supercritical Fluid Solvents

Yoon-Seok Jang and Hun-Soo Byun*



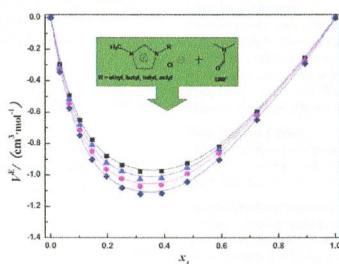
Density Measurement and PC-SAFT/tPC-PSAFT Modeling of the CO₂ + H₂O System over a Wide Temperature Range

Yongchen Song, Weiwei Jian, Yi Zhang,* Mingjun Yang, Jiafei Zhao, Weiguo Liu, Yu Liu, and Yong Shen



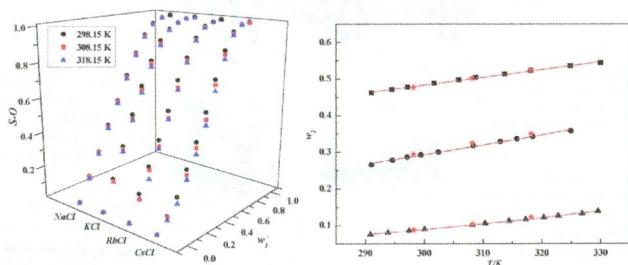
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Xiao-Jing Yan, Shu-Ni Li,* Quan-Guo Zhai, Yu-Cheng Jiang, and Man-Cheng Hu*



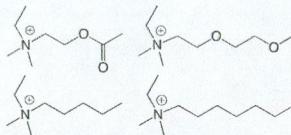
Solid–Liquid Equilibrium (SLE) of the *N,N*-Dimethylacetamide (DMA) + MCl (M = Na, K, Rb, and Cs) + Water Ternary Systems at Multiple Temperatures

Dandan Zhao, Shu'ni Li,* Quanguo Zhai, Yucheng Jiang, and Mancheng Hu*



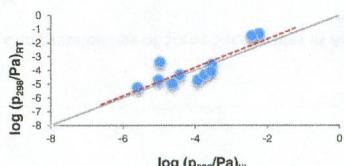
CO₂ Solubilities in Ammonium Bis(trifluoromethanesulfonyl)amide Ionic Liquids: Effects of Ester and Ether Groups

Takashi Makino,* Mitsuhiro Kanakubo,* and Tatsuya Umecky



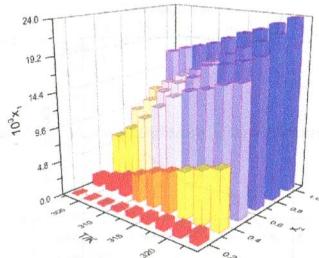
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Sandra Brommer, Liisa M. Jantunen,* Terry F. Bidleman, Stuart Harrad, and Miriam L. Diamond



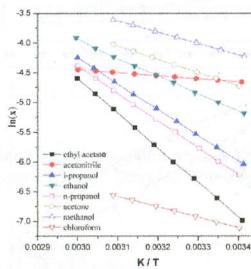
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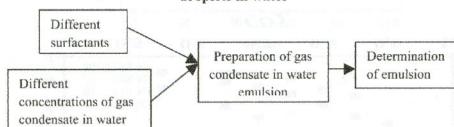
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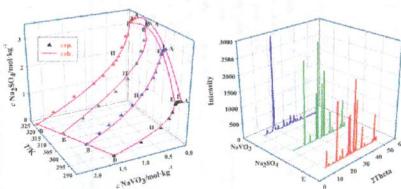
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Hossein Esmaeili, Feridun Esmaeilzadeh,* and Dariush Mowlana

Effect of surfactant on stability and size distribution of gas condensate droplets in water

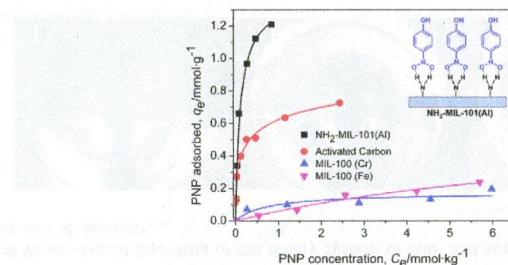


Investigations on the Solubility, Density, and Viscosity in the $\text{NaVO}_3 + \text{Na}_2\text{SO}_4 + \text{H}_2\text{O}$ System from 293.15 K to 323.15 K
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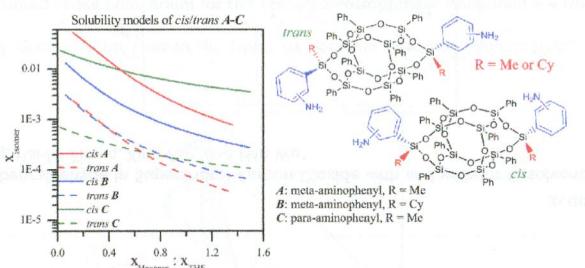
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Baojian Liu,* Fan Yang, Yuanxing Zou, and Yong Peng



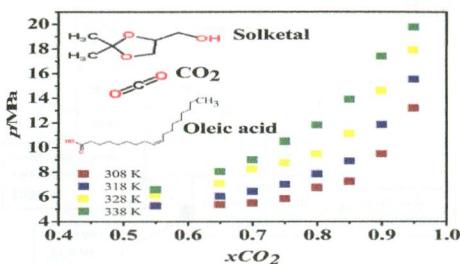
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Beth W. Schoen, Carl T. Lira, and Andre Lee*



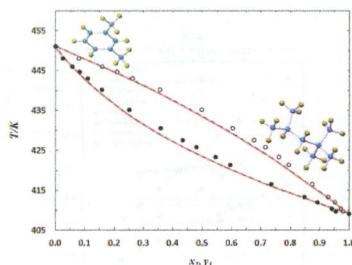
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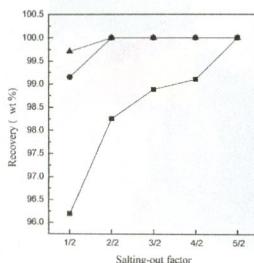
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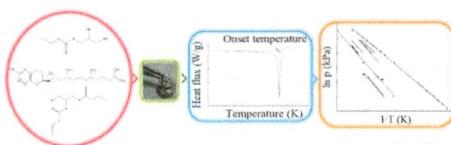
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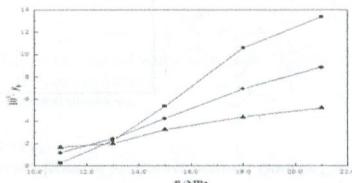
Boiling Points of Short-Chain Partial Acylglycerols and Tocopherols at Low Pressures by the Differential Scanning Calorimetry Technique

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Solubility of 4-Hydroxybenzaldehyde in Supercritical Carbon Dioxide with and without Cosolvents

Jun-su Jin,* Yi-wei Wang, Hai-fei Zhang, Xing Fan, and Hao Wu*



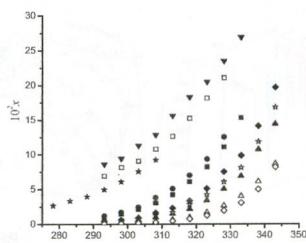
Determination of Vapor–Liquid Equilibria of the Binary System of Sub- and Supercritical CO_2 and Diethyl Fumarate

D. Čuček, Ž. Knez, and M. Škerget*



Solid–Liquid Phase Equilibria of N,N' -[1,3-Phenylenebis(methylene)]bis(phosphoramidic acid) P,P,P',P' -Tetraphenyl Ester in 10 Pure Solvents

Chao-Jun Du, Li-Sheng Wang,* and Guo-Min Yu



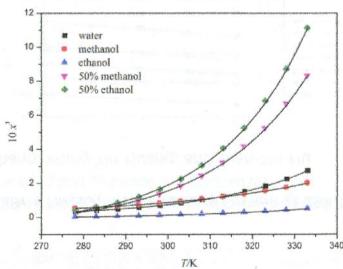
Reactive Extraction of Pyridine-2-carboxylic Acid (Picolinic Acid) Using Nontoxic Extractant and Diluent Systems
Dipaloy Datta and Sushil Kumar*

$$\text{Langmuir Isotherm: } q_e = \frac{Q_0 K_L C_e}{1 + K_L C_e}$$

$$\text{Freundlich Isotherm: } q_e = K_F C_e^{\frac{1}{n}} \quad \text{Temkin Isotherm: } q_e = B_T \log K_T C_e$$

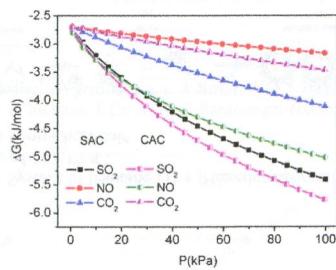
Solubility of Aspartame in Water, Methanol, Ethanol and Different Binary Mixtures in the Temperature Range of (278.15 to 333.15) K

Yixin Leng* and Huichen Qi



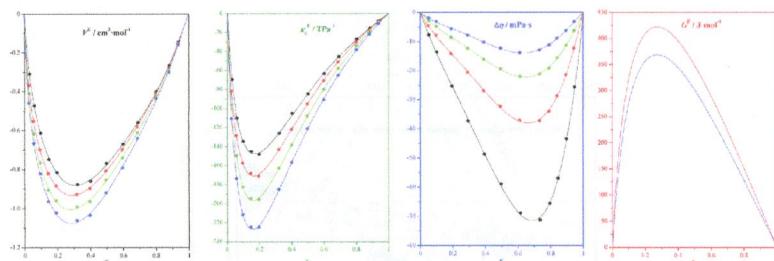
Adsorption of SO₂, NO, and CO₂ on Activated Carbons: Equilibrium and Thermodynamics

Honghong Yi, Zhixiang Wang, Haiyan Liu, Xiaolong Tang,* Ding Ma, Shunzheng Zhao, Bowen Zhang, Fengyu Gao, and Yanran Zuo

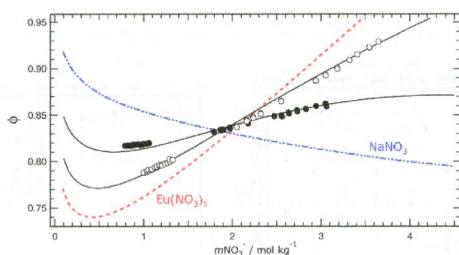


Thermophysical Properties of the Binary Mixture 1-Propylpyridinium Tetrafluoroborate with Methanol

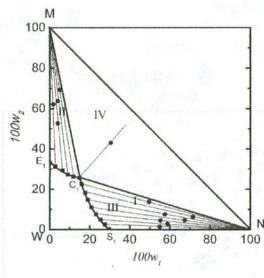
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**Isopiestic Determination of the Osmotic Coefficients of $\text{NaNO}_3 + \text{Eu}(\text{NO}_3)_3 + \text{H}_2\text{O}$ at 298.15 K and Representation with an Extended Ion-Interaction (Pitzer) Model**

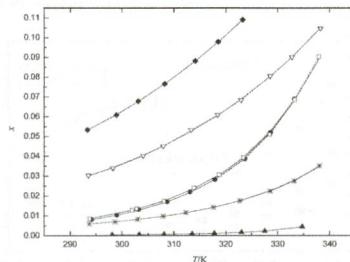
Peter R. Zalupski,* Rocklan McDowell, and Simon L. Clegg*

**Measurement of Solid–Liquid Phase Equilibrium for the Ternary 3-Nitrophthalic Anhydride + 4-Nitrophthalic Anhydride + 1,4-Dioxane System**

Fangli Qiu, Jianguo Yang, Guobo Huang, Huanan Hu, Siying Yu, Hongkun Zhao,* and Rongrong Li*

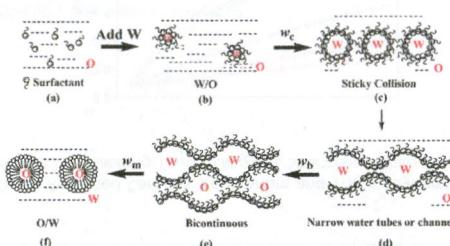


Experimental Measurements and Modeling of the Solubility of Aceclofenac in Six Pure Solvents from (293.35 to 338.25) K
 Jin-Qiang Liu,* Ye Wang, Hui Tang, Sha Wu, Yao-Yao Li, Li-Yue Zhang, Qiao-Yun Bai, and Xun Liu



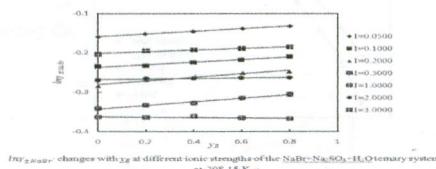
Phase Diagrams for the Pseudoternary System of [Hexane (1) + [Hexadecyltrimethylammonium Bromide (2) + Butan-1-ol (22)]] (2) + Water (3)] at a Temperature of 303.15 K

Xing-Bo Chen, Na Wang, Yahui Ma, and Tong-Chun Bai*



Mean Activity Coefficients of NaBr in NaBr + Na₂SO₄ + H₂O Ternary System at 298.15 K Determined by Potential Difference Measurements

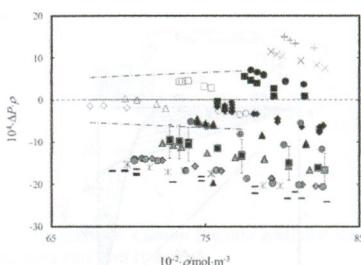
Si-Yao Zhong, Shi-Hua Sang,* Jun-Jie Zhang, Wang-Yin Huang, and Juan-Xin Hu



Influence of ionic strength on $\log f_{\text{NaBr}}^{298.15\text{ K}}$ changes with x_{NaBr} at different ionic strengths of the $\text{NaBr}-\text{Na}_2\text{SO}_4-\text{H}_2\text{O}$ ternary system at 298.15 K.

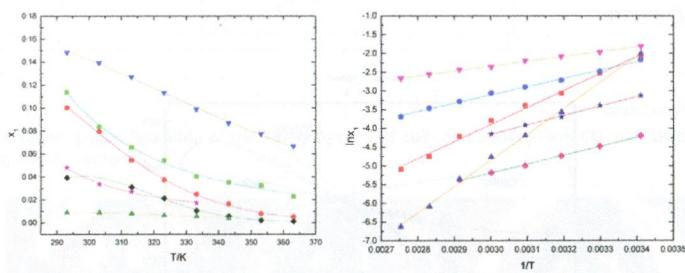
Determination of the Relative Permittivity, ϵ_r , of Octane at Temperatures between (303 and 393) K and Pressures below 25 MPa with a Concentric Cylinder Capacitor at a Frequency of 1 kHz

Laurent Pirolli, Anthony R. H. Goodwin,* Kenneth N. Marsh, and Eric F. May



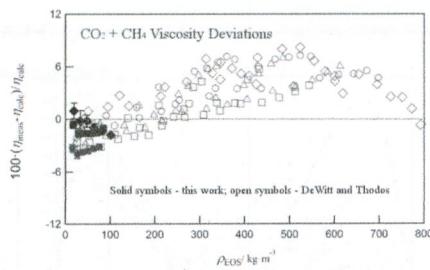
Determination and Correlation of Solubilities of Lithium Bis(oxalate)borate in Six Different Solvents from (293.15 to 363.15) K

Fuchun Chen, Jinhua Hu, Zhengyu Chen, Zhenyu Yang,* and Ningyu Gu*



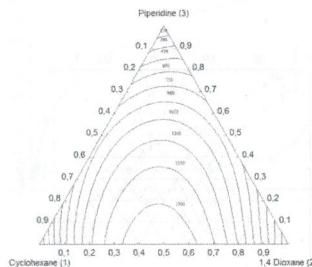
Improved Methods for Gas Mixture Viscosity Using a Vibrating Wire Clamped at Both Ends

Clayton R. Locke, Paul L. Stanwick, Thomas J. Hughes, Austin Kisielew, Anthony R. H. Goodwin, Kenneth N. Marsh, and Eric F. May*



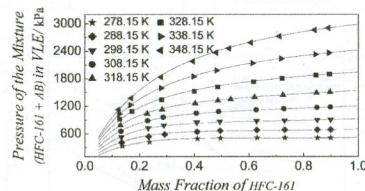
Experimental Measurements and Correlations of Excess Molar Enthalpies for the Binary and Ternary Mixtures of (Cyclohexane, 1,4-Dioxane and Piperidine) or (Cyclohexane, Morpholine and Piperidine) at 308.15 K and Atmospheric Pressure

Farid Brahim Belaribi,* Ghénima Boukais-Belaribi, Atika Dahmoun, Aomar Dahmani, Amir H. Mohammadi, and Dominique Richon



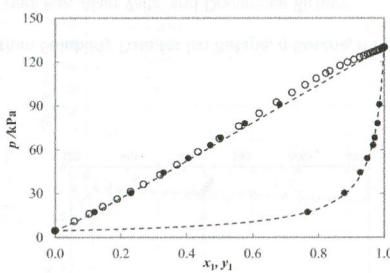
Solubility and Miscibility for the Mixture of (Ethyl Fluoride + Alkylbenzene Oil)

Lingyun Fang, Zanjun Gao, Xiaoyu Wang, Jun Lei, Xiaohong Han,* and Guangming Chen

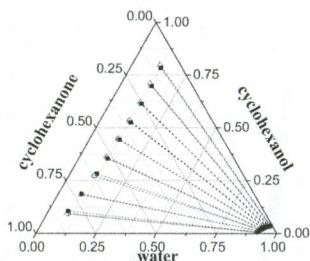


Vapor–Liquid Equilibrium Data for 1-Methyl-2-Pyrrolidone + (1-Butanol or 1-Hexene or Water) Binary Mixtures

Ranjeetha Hirawan, Sumit Sinha, Samuel A. Iwarere, J. David Raal, Paramespri Naidoo, and Deresh Ramjugernath*

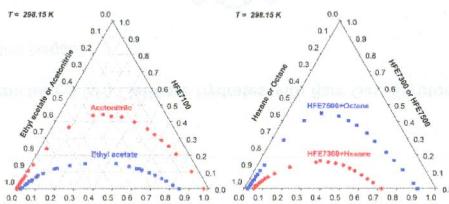


Measurements and Correlation of Liquid–Liquid Equilibria for the Ternary System Water + Cyclohexanol + Cyclohexanone
Xing Gong, Qinbo Wang,* Fuqiong Lei, and Binwei Shen

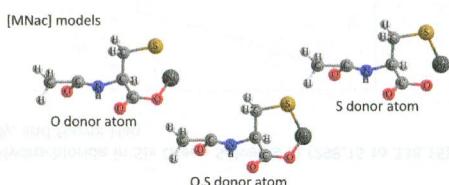


Liquid–Liquid Equilibria for the Ternary Systems of FC3283 + HFE7300 + Hexane, FC3283 + HFE7500 + Octane, and FC72 + HFE7100 + (Acetonitrile or Ethyl Acetate) at 273.15 K, 298.15 K, and 313.15 K

Sang Young Lim, Tae Gyu Lee, Kwang Ho Song,* and Jaehoon Choe

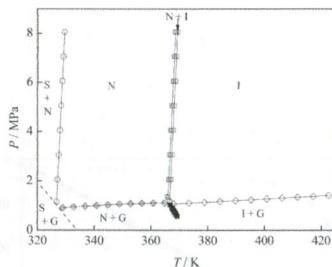


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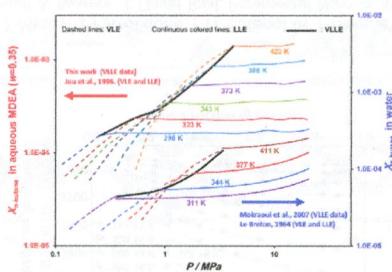


Phase Behavior of Liquid Crystal + CO₂ Mixtures

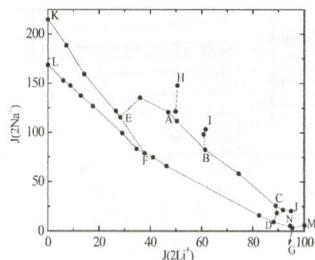
Mariëtte de Groen, Bo C. Ramaker, Thijs J. H. Vlugt, and Theo W. de Loos*

**New Vapor-Liquid-Liquid Equilibrium Solubility Data for iso-Butane, n-Butane, n-Pentane, and n-Hexane in Alkanolamine Aqueous Solutions**

Salim Mokraoui,* Mohamed Kamel Hadj-Kali, Alain Valtz, and Dominique Richon

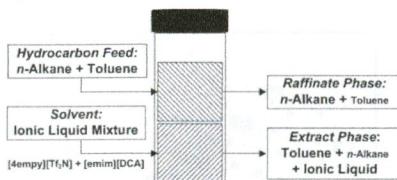
**Solid-Liquid Metastable Phase Equilibria in the Five-Component System (Li + Na + K + Cl + SO₄ + H₂O) at 308.15 K**

Yuanhui Liu, Yafei Guo, Xiaoping Yu, Shiqiang Wang, and Tianlong Deng*

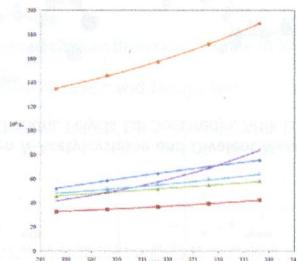


Liquid-Liquid Extraction of Toluene from *n*-Alkanes using $\{[4\text{empy}][\text{Tf}_2\text{N}] + [\text{emim}][\text{DCA}]\}$ Ionic Liquid Mixtures

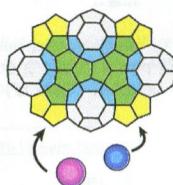
Marcos Larriba, Pablo Navarro, Julián García,* and Francisco Rodríguez

**Solubility of Metoclopramide Hydrochloride in Six Green Solvents at (298.15 to 338.15) K**

Faiyaz Shakeel,* Gamal A. Shazly, and Nazrul Haq

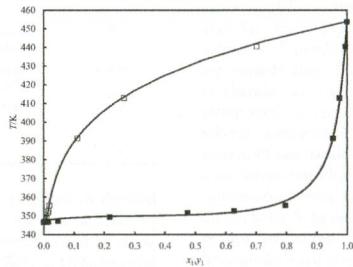
**Crystal Phase Boundaries of Structure-H (sH) Clathrate Hydrates with Rare Gas (Krypton and Xenon) and Bromide Large Molecule Guest Substances**

Yusuke Jin,* Masato Kida, and Jiro Nagao*



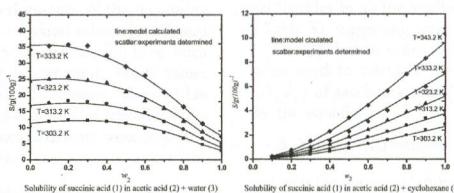
Vapor-Liquid Equilibrium Data for Binary Systems of *n*-Dodecane + {Propan-1-ol, Butan-1-ol, 2-Methylpropan-1-ol} at 40 kPa

Zoubir Tebbal, Peterson Thokozani Ngema, Caleb Narasigadu, Latifa Negadi,* and Deresh Ramjugernath



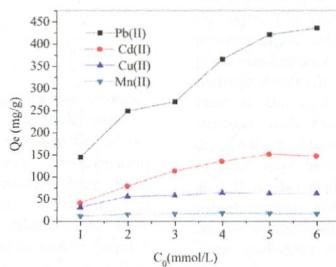
Solubilities of Succinic Acid in Acetic Acid + Water Mixtures and Acetic Acid + Cyclohexane Mixtures

Fuqiong Lei, Qinbo Wang,* Xing Gong, Binwei Shen, Wenming Zhang, and Qing Han



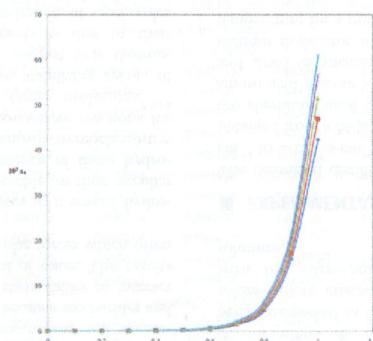
Preparation of Graphene-Oxide/Polyamidoamine Dendrimers and Their Adsorption Properties toward Some Heavy Metal Ions

Fan Zhang,* Bo Wang, Shengfu He, and Ruilin Man



Solubility of *N*-(4-Chlorophenyl)-2-(pyridin-4-ylcarbonyl)hydrazinecarbothioamide (Isoniazid Analogue) in Transcutol + Water Cosolvent Mixtures at (298.15 to 338.15) K

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The solubility of the title compound in pure water at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure water at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure Transcutol at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure Transcutol at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure ethanol at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure ethanol at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure acetone at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure acetone at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure methanol at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure methanol at 338.15 K was found to be 1.5 g/g.

RESULTS AND DISCUSSION

The solubility of the title compound in pure water at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure water at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure Transcutol at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure Transcutol at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure ethanol at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure ethanol at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure acetone at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure acetone at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure methanol at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure methanol at 338.15 K was found to be 1.5 g/g.

CONCLUSION

The solubility of the title compound in pure water at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure water at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure Transcutol at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure Transcutol at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure ethanol at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure ethanol at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure acetone at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure acetone at 338.15 K was found to be 1.5 g/g. The solubility of the title compound in pure methanol at 298.15 K was found to be 0.5 g/g. The solubility increased with increasing temperature. The solubility in pure methanol at 338.15 K was found to be 1.5 g/g.

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