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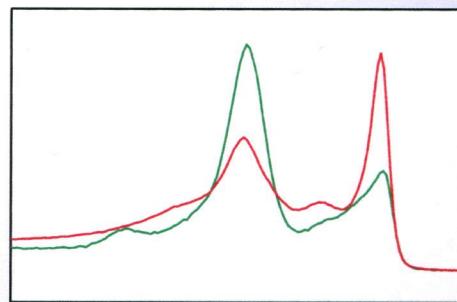
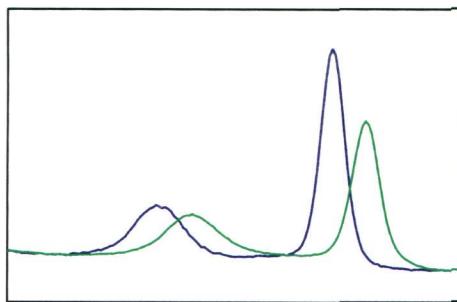
# THE JOURNAL OF PHYSICAL CHEMISTRY C

TiO<sub>2</sub> Doping Enhances  
the Catalytic Activity  
of RuO<sub>2</sub> through  
Charge Transfer  
(see page 5A)



High Activity

Low Activity



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INTERFACES, NANOMATERIALS, AND HARD MATTER



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# THE JOURNAL OF PHYSICAL CHEMISTRY

**ON THE COVER:** TiO<sub>2</sub> doping enhances the catalytic activity of RuO<sub>2</sub> through charge transfer. The optical image displays DSA-type electrode where pure RuO<sub>2</sub> (right) and TiO<sub>2</sub>-doped RuO<sub>2</sub> (left) coatings are separated by TiO<sub>2</sub>. The scanning electrochemical microscopy shows the UME tip response from the OER activity on the substrate electrode, indicating a higher electrochemical activity for the TiO<sub>2</sub>-doped RuO<sub>2</sub> even though the Ru content in the mixed (Ru<sub>1-x</sub>:Ti<sub>x</sub>)O<sub>2</sub> coating is considerably lower compared with pure RuO<sub>2</sub> coating. X-ray photoelectron spectroscopy suggests that the electrocatalytic enhancement for OER on the mixed (Ru<sub>1-x</sub>:Ti<sub>x</sub>)O<sub>2</sub> coating is promoted through a charge transfer from the RuO<sub>2</sub> to the TiO<sub>2</sub>, which provides new and more reactive sites designated as activated RuO<sub>2</sub><sup>δ+</sup>. See page 6126.

## Articles

### Energy Conversion and Storage; Energy and Charge Transport

6013

[dx.doi.org/10.1021/jp309724q](https://doi.org/10.1021/jp309724q)

**Coralline Glassy Lithium Phosphate-Coated LiFePO<sub>4</sub> Cathodes with Improved Power Capability for Lithium Ion Batteries**  
Guoqiang Tan, Feng Wu, Li Li,\* Renjie Chen,\* and Shi Chen

6022

[dx.doi.org/10.1021/jp310555b](https://doi.org/10.1021/jp310555b)

**Synthesis and Electrochemical and Computational Analysis of Two New Families of Thiophene-Carbonyl Molecules**  
Weidong Zhou, Kenneth Hernández-Burgos, Stephen E. Burkhardt, Hualei Qian, and Héctor D. Abruna\*

6033

[dx.doi.org/10.1021/jp3112556](https://doi.org/10.1021/jp3112556)

**High-Pressure Electrical-Transport Properties of SnS: Experimental and Theoretical Approaches**

Feng Ke, Jie Yang, Cailong Liu, Qinglin Wang, Yuqiang Li, Junkai Zhang, Lei Wu, Xin Zhang, Yonghao Han, Baojia Wu,\* Yanzhang Ma, and Chunxiao Gao\*

6039

[dx.doi.org/10.1021/jp311336v](https://doi.org/10.1021/jp311336v)

**Charge Transfer States in Merocyanine Neat Films and Its Blends with [6,6]-Phenyl-C<sub>6</sub>-butyric Acid Methyl Ester**  
Domantas Peckus, Andrius Devižis, Ramūnas Augulis, Steven Graf, Dirk Hertel,\* Klaus Meerholz, and Vidmantas Gulbinas\*

6049

[dx.doi.org/10.1021/jp3118067](https://doi.org/10.1021/jp3118067)

**Big Bandgap in Highly Reduced Graphene Oxides**

Ke-Yan Lian, Yong-Fei Ji, Xiao-Fei Li, Ming-Xing Jin, Da-Jun Ding, and Yi Luo\*

6055

[dx.doi.org/10.1021/jp311972f](https://doi.org/10.1021/jp311972f)

**Functionalized Graphitic Carbon Nitride for Efficient Energy Storage**

Menghao Wu,\* Qian Wang, Qiang Sun, and Puru Jena

6060

S

**Reversible Hydrogenation Studies of NaBH<sub>4</sub> Milled with Ni-Containing Additives**

Terry D. Humphries,\* Georgios N. Kalantopoulos, Isabel Llamas-Jansa, Jørn Eirik Olsen, and Bjørn C. Hauback

dx.doi.org/10.1021/jp312105w

dx.doi.org/10.1021/jp308941g

6066

S

**Role of Adsorption Structures of Zn-Porphyrin on TiO<sub>2</sub> in Dye-Sensitized Solar Cells Studied by Sum Frequency Generation Vibrational Spectroscopy and Ultrafast Spectroscopy**

Shen Ye,\* Arunkumar Kathiravan, Hironobu Hayashi, Yujin Tong, Yingyot Infahsaeng, Pavel Chabera, Torbjörn Pascher, Arkady P. Yartsev, Seiji Isoda, Hiroshi Imahori,\* and Villy Sundström\*

dx.doi.org/10.1021/jp400336r

dx.doi.org/10.1021/jp309394p

6081.

**Local Structural Modifications versus Transport Properties in AgI-Doped Silver–Borate Glasses: A Detailed X-ray Absorption Investigation**

Andrea Sanson,\* Cristina Armellini, Rolly Grisenti, and Paolo Fornasini

dx.doi.org/10.1021/jp400735n

dx.doi.org/10.1021/jp309401q

**Surfaces, Interfaces, Porous Materials, and Catalysis**

6088

**Viscosity of Ultrathin Water Films Confined between Aluminol Surfaces of Kaolinite: Ab Initio Simulations**

Peter J. Feibelman\*

dx.doi.org/10.1021/jp312152h

dx.doi.org/10.1021/jp309483p

6096

**Viscosity of Nanoconfined Water between Hydroxyl Basal Surfaces of Kaolinite: Classical Simulation Results**

Neil R. Haria, Gary S. Grest, and Christian D. Lorenz\*

dx.doi.org/10.1021/jp312181u

dx.doi.org/10.1021/jp3102496

6105

S

**Role of Interfacial Energy and Crystallographic Orientation on the Mechanism of the ZnO + Al<sub>2</sub>O<sub>3</sub> → ZnAl<sub>2</sub>O<sub>4</sub> Solid-State Reaction: I. Reactivity of Films Deposited onto the Sapphire (110) and (012) Faces**

Sonia Pin, Marco Suardelli, Francesco D'Acapito, Giorgio Spinolo, Michele Zema, Serena C. Tarantino, and Paolo Ghigna\*

dx.doi.org/10.1021/jp3124956

dx.doi.org/10.1021/jp310341z

6113

**Role of Interfacial Energy and Crystallographic Orientation on the Mechanism of the ZnO + Al<sub>2</sub>O<sub>3</sub> → ZnAl<sub>2</sub>O<sub>4</sub> Solid-State Reaction: II. Reactivity of Films Deposited onto the Sapphire (001) Face**

Sonia Pin, Marco Suardelli, Francesco D'Acapito, Giorgio Spinolo, Michele Zema, Serena C. Tarantino, Luisa Barba, and Paolo Ghigna\*

dx.doi.org/10.1021/jp312517w

dx.doi.org/10.1021/jp311439p

6120

S

**Formation of a Single Gold Nanoparticle on a Nanometer-Sized Electrode and Its Electrochemical Behaviors**

Peng Sun,\* Fei Li, Cheng Yang, Tong Sun, Ismail Kady, Benjamin Hunt, and Jian Zhuang

dx.doi.org/10.1021/jp308501j

dx.doi.org/10.1021/jp311832d

6126

S

**The Role of TiO<sub>2</sub> Doping on RuO<sub>2</sub>-Coated Electrodes for the Water Oxidation Reaction**

Lars-Åke Näslund,\* Carlos M. Sánchez-Sánchez, Árni S. Ingason, Joakim Bäckström, Enrique Herrero, Johanna Rosen, and Susanne Holmin

dx.doi.org/10.1021/jp3119154

6136

S

**Microkinetic Simulation of Temperature-Programmed Desorption**

Chia-Ching Wang, Jyun-Yi Wu, and Jyh-Chiang Jiang\*

dx.doi.org/10.1021/jp309394p

6143

S

**Methanol Steam Reforming over Indium-Promoted Pt/Al<sub>2</sub>O<sub>3</sub> Catalyst: Nature of the Active Surface**

Roland L. Barbosa, Vasiliki Papaefthimiou, Yeuk T. Law, Detre Teschner, Michael Hävecker, Axel Knop-Gericke, Ralf Zapf, Gunther Kolb, Robert Schlögl, and Spyridon Zafeiratos\*

dx.doi.org/10.1021/jp309401q

6151

S

**Properties of Adsorbed Bovine Serum Albumin and Fibrinogen on Self-Assembled Monolayers**

Latit M. Pandey, Sudip K. Pattanayek,\* and Didier Delabougline

dx.doi.org/10.1021/jp309483p

6161

**Sulfur- and Oxygen-Induced Alterations of the Iron (001) Surface Magnetism and Work Function: A Theoretical Study**

Håkan Wilhelm Hugosson,\* Weimin Cao, Seshadri Seetharaman, and Anna Delin

dx.doi.org/10.1021/jp3102496

6172

**Water Adsorption and Decomposition on N/V-Doped Anatase TiO<sub>2</sub> (101) Surfaces**

Zongyan Zhao, Zhaosheng Li,\* and Zhigang Zou\*

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6185

S

**Honeycomb Porous Films Prepared from Porphyrin-Cored Star Polymers: Submicrometer Pores Induced by Transition of Monolayer into Multilayer Structures**

Liang-Wei Zhu, Ling-Shu Wan,\* Jing Jin, and Zhi-Kang Xu

dx.doi.org/10.1021/jp311439p

6195

S

**High-Speed Imaging of Freezing Drops: Still No Preference for the Contact Line**

Colin Gurganus, Alexander B. Kostinski, and Raymond A. Shaw\*

dx.doi.org/10.1021/jp311832d

6201

S

**Surface Chemistry of Nanosized Hydrated Ferric Oxide Encapsulated Inside Porous Polymer: Modeling and Experimental Studies**

Guangze Nie, Bingcai Pan,\* Shujuan Zhang, and Bingjun Pan

dx.doi.org/10.1021/jp3119154

6210

[dx.doi.org/10.1021/jp311924q](https://doi.org/10.1021/jp311924q)

Analyzing the Influence of H<sub>3</sub>PO<sub>4</sub> as Catalyst Poison in High Temperature PEM Fuel Cells Using *in-operando* X-ray

Absorption Spectroscopy

Sebastian Kaserer, Keegan M. Caldwell, David E. Ramaker, and Christina Roth\*

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6218



[dx.doi.org/10.1021/jp312444s](https://doi.org/10.1021/jp312444s)

Synthesis of the Catalytically Active Mn<sub>3</sub>O<sub>4</sub> Spinel and Its Thermal Properties

Zhen-Yu Tian,\* Patrick Mountaprbeme Kouotou, Naoufal Bahlawane, and Patrick Hervé Tchoua Ngamou

[dx.doi.org/10.1021/jp400507w](https://doi.org/10.1021/jp400507w)

6225



[dx.doi.org/10.1021/jp312548g](https://doi.org/10.1021/jp312548g)

Quantum-Chemical Investigation of Hydrocarbon Oxidative Dehydrogenation over Spin-Active Carbon Catalyst Clusters

Oleksiy V. Khavryuchenko,\* Benjamin Frank, Annette Trunschke,\* Klaus Hermann, and Robert Schlögl

[dx.doi.org/10.1021/jp400698z](https://doi.org/10.1021/jp400698z)

6235

Gas Flows near Solids Coated with Thin Water Films

Dongjin Seo, Dean Mastropietro, and William A. Ducker\*

[dx.doi.org/10.1021/jp312568y](https://doi.org/10.1021/jp312568y)

[dx.doi.org/10.1021/jp4007763](https://doi.org/10.1021/jp4007763)

6245

Hofmeister Effects on Cation Exchange Equilibrium: Quantification of Ion Exchange Selectivity

Xinmin Liu, Hang Li,\* Wei Du, Rui Tian, Rui Li, and Xianjun Jiang

[dx.doi.org/10.1021/jp312682u](https://doi.org/10.1021/jp312682u)

[dx.doi.org/10.1021/jp400838j](https://doi.org/10.1021/jp400838j)

6252

Multifunctionalized Ordered Mesoporous Carbon as an Efficient and Stable Solid Acid Catalyst for Biodiesel Preparation

Binbin Chang, Jie Fu, Yanlong Tian, and Xiaoping Dong\*

[dx.doi.org/10.1021/jp312820g](https://doi.org/10.1021/jp312820g)

[dx.doi.org/10.1021/jp400902f](https://doi.org/10.1021/jp400902f)

6259



[dx.doi.org/10.1021/jp4000297](https://doi.org/10.1021/jp4000297)

Dealloying of Cobalt from CuCo Nanoparticles under Syngas Exposure

Sophie Carenco, Anders Tuxen, Mahati Chintapalli, Elizbieta Pach, Carlos Escudero, Trevor D. Ewers, Peng Jiang, Ferenc Borondics, Geoff Thornton, A. Paul Alivisatos, Hendrik Bluhm, Jinghua Guo, and Miquel Salmeron\*

6325

Energetics of Adsorbed CH<sub>3</sub> and CH on Pt(111) by Calorimetry: Dissociative Adsorption of CH<sub>3</sub>

Eric M. Karp, Trent L. Silbaugh, and Charles T. Campbell\*

[dx.doi.org/10.1021/jp4009459](https://doi.org/10.1021/jp4009459)

6267

Chemical Structure of Oxidized Multilayer Epitaxial Graphene: A Density Functional Theory Study

Si Zhou, Suenne Kim, and Angelo Bongiorno\*

[dx.doi.org/10.1021/jp400128t](https://doi.org/10.1021/jp400128t)

[dx.doi.org/10.1021/jp400971h](https://doi.org/10.1021/jp400971h)

6275



[dx.doi.org/10.1021/jp400235y](https://doi.org/10.1021/jp400235y)

Understanding the Equilibria of Thio Compounds Adsorbed on Gold by Surface-Enhanced Raman Scattering and Density Functional Theory Calculations

Tércio de F. Paulo, Rômulo A. Ando, Izaura C. N. Diógenes,\* and Marcia L. A. Temperini\*

6337

Electronically Nonadiabatic Processes in the Interaction of H with a Au Surface Revealed Using MIM Junctions: The Temperature Dependence

Beate Schindler, Detlef Diesing, and Eckart Hasselbrink\*

[dx.doi.org/10.1021/jp4009459](https://doi.org/10.1021/jp4009459)

6346

Reciprocating Motion of a Self-Propelled Object on a Molecular Layer with a Local Minimum and a Local Maximum Isotherm

Satoshi Nakata,\* Tatsuya Miyaji, Tomoaki Ueda, Taisuke Sato, Yumihiro S. Ikura, Shunsuke Izumi, and Masaharu Nagayama

[dx.doi.org/10.1021/jp400971h](https://doi.org/10.1021/jp400971h)

6353

Role of Magnetism in Catalysis: RuO<sub>2</sub> (110) Surface

E. Torun, C. M. Fang, G. A. de Wijs, and R. A. de Groot\*

[dx.doi.org/10.1021/jp4020367](https://doi.org/10.1021/jp4020367)

6358  dx.doi.org/10.1021/jp402100v

**Surface Structure and Reactivity of Anatase TiO<sub>2</sub> Crystals with Dominant {001} Facets**  
Sencer Selçuk and Annabella Selloni\*

## Plasmonics, Optical Materials, and Hard Matter

6363  dx.doi.org/10.1021/jp311467b

**Multilayer Magnetic Composite Particles with Functional Polymer Brushes as Stabilizers for Gold Nanocolloids and Their Recyclable Catalysis**  
Bin Liu, Dongwei Zhang, Jianchao Wang, Cheng Chen, Xinlin Yang,\* and Chenxi Li

6373  dx.doi.org/10.1021/jp401372k

**Optical Resonances in Short-Range Ordered Nanoholes in Ultrathin Aluminum/Aluminum Nitride Multilayers**  
Yuichiro Ikenoya, Masahiro Susa, Ji Shi, Yoshio Nakamura, Andreas B. Dahlin, and Takumi Sannomiya\*

6383  dx.doi.org/10.1021/jp3121963

**Optical Properties of a Particle above a Dielectric Interface: Cross Sections, Benchmark Calculations, and Analysis of the Intrinsic Substrate Effects**

Jean Lermé,\* Christophe Bonnet, Michel Broyer, Emmanuel Cottancin, Delphine Manchon, and Michel Pellarin

6399  dx.doi.org/10.1021/jp401861c

**NMR, ESR, and Luminescence Characterization of Bismuth Embedded Zeolites Y**  
Hong-Tao Sun,\* Yoshio Sakka, Naoto Shirahata, Yoshitaka Matsushita, Kenzo Deguchi, and Tadashi Shimizu

## Physical Processes in Nanomaterials and Nanostructures

6409 dx.doi.org/10.1021/jp3101046

**Proton Binding and Ion Exchange at the Akaganéite/Water Interface**  
Philipp A. Kozin and Jean-François Boily\*

6420 dx.doi.org/10.1021/jp310463k

**Electrical Spin Switch in Hydrogenated Multilayer Graphene**  
Elton J. G. Santos\*

6426 dx.doi.org/10.1021/jp311471f

**Giant Magnetoresistive Phosphoric Acid Doped Polyaniline–Silica Nanocomposites**  
Hongbo Gu, Jiang Guo, Xi Zhang, Qingliang He, Yudong Huang, Henry A. Colorado, Neel Haldolaarachchige, Huolin Xin, David P. Young, Suying Wei,\* and Zhanhu Guo\*

6437  dx.doi.org/10.1021/jp311948c

**Structure of Tm@C<sub>62</sub>(I) Metallofullerene by Single-Crystal X-ray Diffraction Using the 1:2 Co-Crystal with Octaethylporphyrin Nickel (Ni(OEP))**  
Yuki Sado, Shinobu Aoyagi, Ryo Kitaura, Yasumitsu Miyata, Eiji Nishibori, Hiroshi Sawa, Kunihisa Sugimoto, and Hisanori Shinohara\*

6443 dx.doi.org/10.1021/jp312080v

**Optical and Magnetic Properties of Zn<sub>1-x</sub>Mn<sub>x</sub>O Nanorods Grown by Chemical Vapor Deposition**  
The-Long Phan and S. C. Yu\*

6454  dx.doi.org/10.1021/jp4000329

**Enhanced Ultraviolet Emission from Mg Doped SnO<sub>2</sub> Nanocrystals at Room Temperature and Its Modulation upon H<sub>2</sub> Annealing**  
Nilesh Mazumder, Dipayan Sen, Subhajit Saha, Uttam Kumar Ghorai, Nirmalya Sankar Das, and Kalyan Kumar Chattopadhyay\*

6462 dx.doi.org/10.1021/jp4002687

**Density of Surface States at CdSe Quantum Dots by Fitting of Temperature-Dependent Surface Photovoltage Transients with Random Walk Simulations**  
Steffen Fengler, Elisabeth Zillner, and Thomas Dittrich\*

6469 dx.doi.org/10.1021/jp400274a

**Substrate Mediation in Vapor Deposition Growth of Layered Chalcogenide Nanoplates: A Case Study of SnSe<sub>2</sub>**  
Liang Huang, Yifei Yu, Chun Li, and Linyou Cao\*

6476 dx.doi.org/10.1021/jp4003774

**Finite-Height Effect on Electron Energy Structure of Lead Salts Nanorods**  
S. V. Gopalov\*