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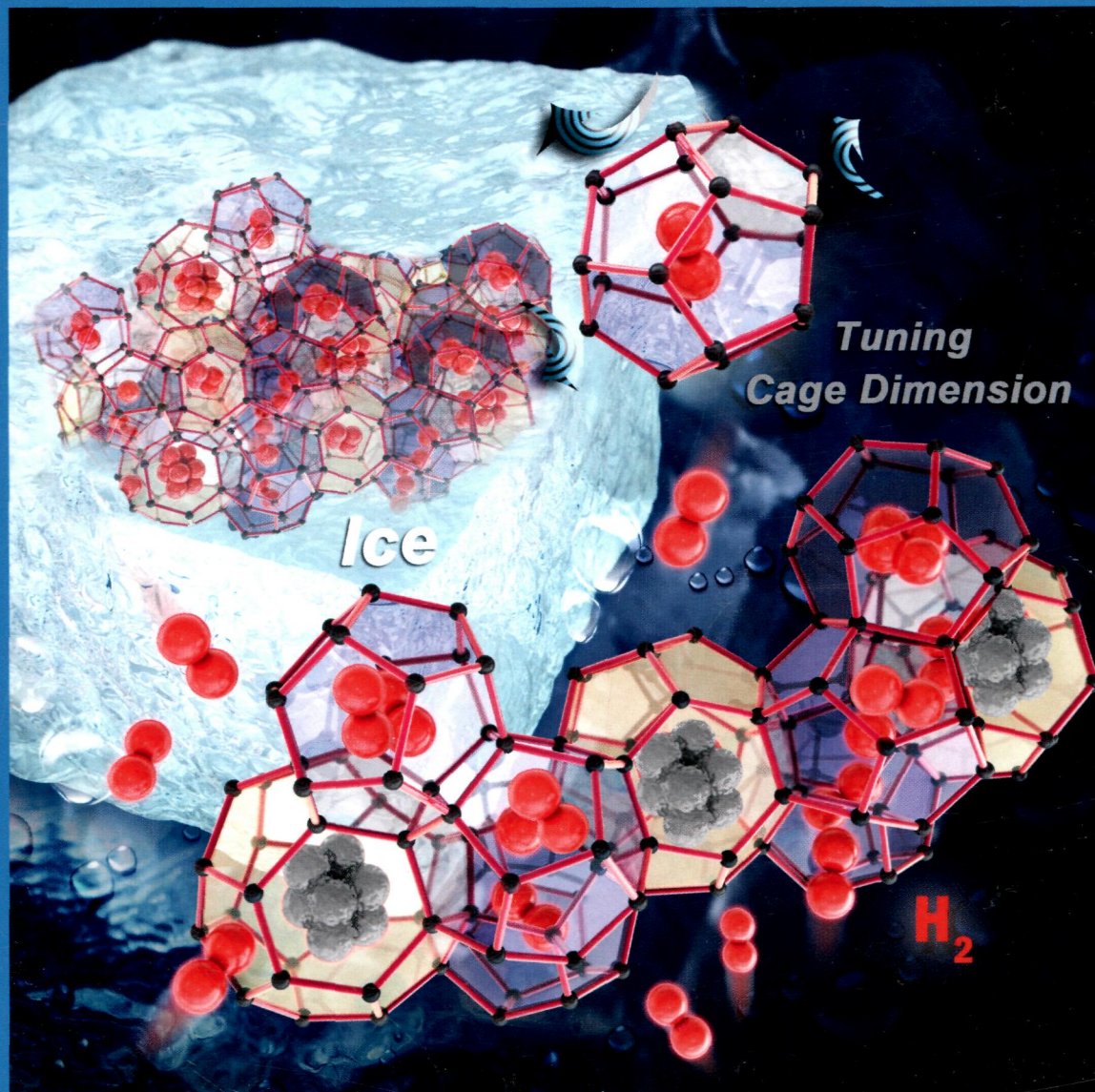
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Tunable Icy Cages for
Improving Hydrogen
Storage Capacity
(see page 5A)

ENERGY CONVERSION AND STORAGE, OPTICAL AND ELECTRONIC DEVICES,
INTERFACES, NANOMATERIALS, AND HARD MATTER



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ON THE COVER: Tunable icy cages for improving hydrogen storage capacity. This study reports on how clathrate hydrate lattices can be tuned to result in multiple occupancy in ubiquitous nano-sized dodecahedral cavities (S^{12}), which are generally known to accept only one hydrogen molecule. Size and population of large guest molecule (LGM) in the hydrate matrix spontaneously control the degree of molecular hydrogen storage by tuning the cage dimensions. Generalization of this new insight to other sI and sII hydrates will raise the accepted hydrogen storage quantity limit in icy materials. See page 3324–3330.

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Energy Conversion and Storage; Energy and Charge Transport

- 2897  [dx.doi.org/10.1021/jp5000057](https://doi.org/10.1021/jp5000057)
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- 2908 [dx.doi.org/10.1021/jp409581n](https://doi.org/10.1021/jp409581n)
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- 2919 [dx.doi.org/10.1021/jp409762m](https://doi.org/10.1021/jp409762m)
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- 2929  [dx.doi.org/10.1021/jp4104273](https://doi.org/10.1021/jp4104273)
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- 2943 [dx.doi.org/10.1021/jp410502s](https://doi.org/10.1021/jp410502s)
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- 3102 dx.doi.org/10.1021/jp4114128
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Additions and Corrections

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Correction to "Thermodynamics of Ionized Monolayers: Surface Manometry on Very Low Density Spread Monolayers of Sodium Octadecyl Sulfate at the Air/Water Interface and Analysis of Ionic Double Layer Contributions to the Isotherms"
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