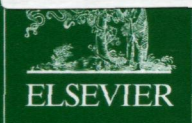


FM
J80/s2s

Volume 207

November 2013

ISSN 0022-4596



JOURNAL OF SOLID STATE CHEMISTRY

Editor

M.G. KANATZIDIS

Associate Editors

S.J. HWANG

J. LI

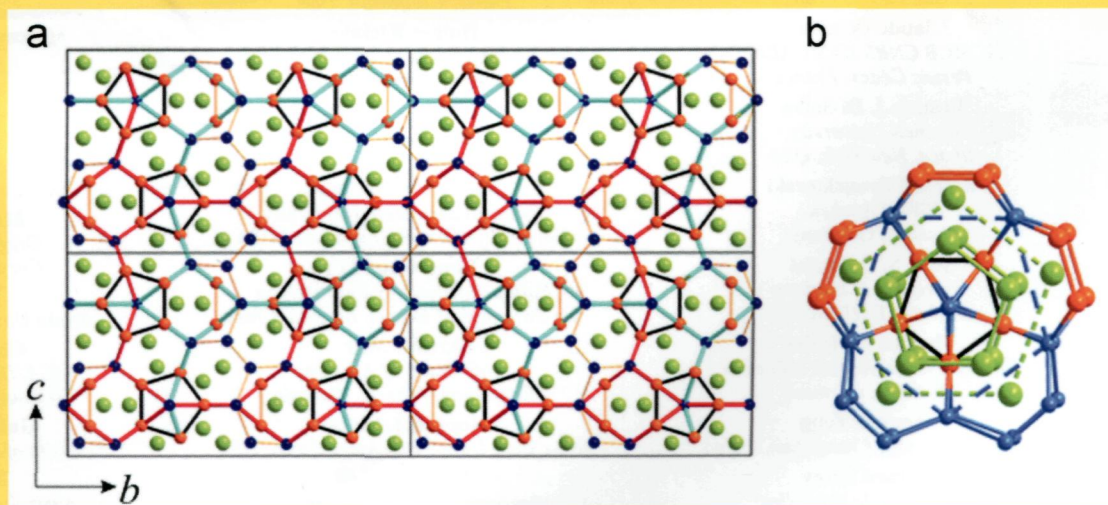
W. TREMEL

S.J. CLARKE

H.-C. ZUR LOYE

IN THIS ISSUE:

$\text{Na}_8\text{Au}_{9.8(4)}\text{Ga}_{7.2}$ and $\text{Na}_{17}\text{Au}_{5.87(2)}\text{Ga}_{46.63}$: The diversity of pseudo 5-fold symmetries in the Na–Au–Ga system



Volodymyr Smetana, John D. Corbett and Gordon J. Miller

Available online at www.sciencedirect.com

ScienceDirect

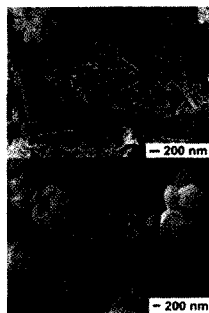
J
S
S
C

Abstracted/indexed in BioEngineering Abstracts, Chemical Abstracts, Coal Abstracts, Current Contents/Physics, Chemical, & Earth Sciences, Engineering Index, Research Alert, SCISEARCH, Science Abstracts, and Science Citation Index. Also covered in the abstract and citation database SCOPUS[®]. Full text available on ScienceDirect[®].

Regular Articles

Morphology and mechanism study for the synthesis of ZrB₂-SiC powders by different methods

Bin Zhao, Yun Zhang, Junping Li, Biyun Yang, Tingyu Wang, Yongzhen Hu, Dongfeng Sun, Ruixing Li, Shu Yin, Zhihai Feng and Tsugio Sato
page 1

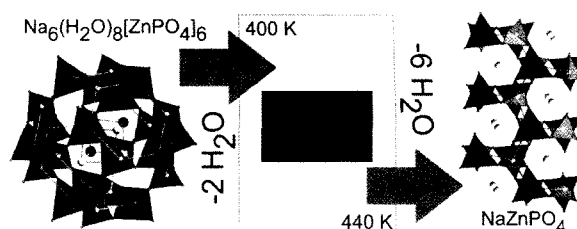


SEM images of the ZrB₂-SiC powders obtained by different methods.

Regular Articles—Continued

Temperature-dependent framework–template interaction of Na₆(H₂O)₈[ZnPO₄]₆ sodalite

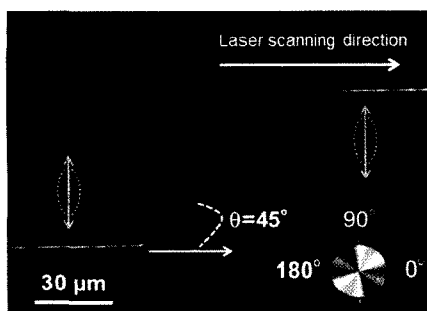
Lars Robben and Thorsten M. Gesing
page 13



The thermal decomposition of Na₆(H₂O)₈[ZnPO₄]₆ sodalite is a two-step process with an initial loss of two water molecules at 400 K, intermediately stabilizing the highly stressed framework. Further heating leads to a subsequent loss of the remaining six water molecules and a breakup of the sodalite framework. A beryllonite-type NaZnPO₄ is the final product of this process.

Birefringence imaging and orientation of laser patterned β-BaB₂O₄ crystals with bending and curved shapes in glass

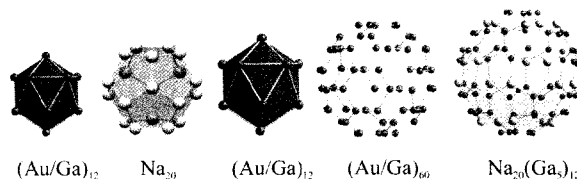
Kazuki Ogawa, Tsuyoshi Honma and Takayuki Komatsu
page 6



This figure shows the birefringence images obtained by the Abrio IM imaging system ($\lambda=546$ nm) for the laser-patterned β -BaB₂O₄ crystal line with the bending angle of 45° in the glass. The relation between the direction of slow axis and color is also shown. It is demonstrated that the formation (crystallization) of highly *c*-axis oriented β -BaB₂O₄ crystals follows along laser scanning direction even if the laser scanning direction changes.

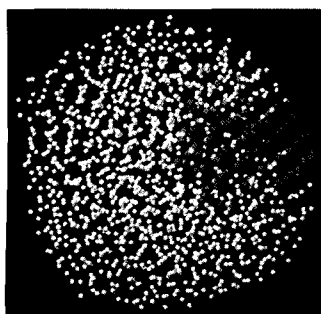
Na₈Au_{9.8(4)}Ga_{7.2} and Na₁₇Au_{5.87(2)}Ga_{46.63}: The diversity of pseudo 5-fold symmetries in the Na–Au–Ga system

Volodymyr Smetana, John D. Corbett and Gordon J. Miller
page 21



Multiply-endohedral Bergman-related clusters in the structure of Na₁₇Au_{5.9(1)}Ga_{46.6}.

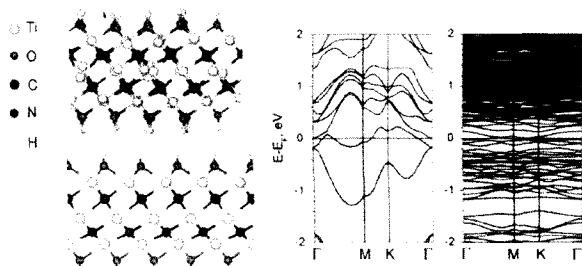
Molecular dynamics studies of the size and temperature dependence of the kinetics of freezing of Fe nanoparticles
Bo Zhao, Jinfan Huang and Lawrence S. Bartell
page 35



Critical nuclei of crystallization.

Two-dimensional titanium carbonitrides and their hydroxylated derivatives: Structural, electronic properties and stability of MXenes $Ti_3C_{2-x}N_x(OH)_2$ from DFTB calculations

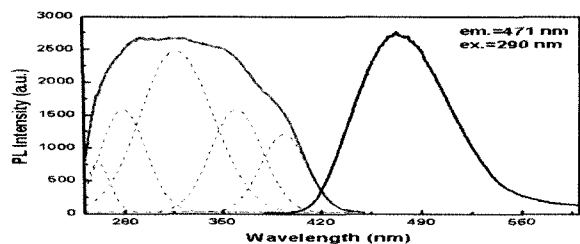
A.N. Enyashin and A.L. Ivanovskii
page 42



The side views of the optimized atomic structures of some examined hydroxylated derivatives of MXene Ti_3CN and their electronic band structures.

Luminescence properties of a blue-emitting phosphor: $(Sr_{1-x}Eu_x)Si_9Al_{19}ON_{31}$ ($0 < x \leq 1$)

Lihong Liu, Rong-Jun Xie, Naoto Hirosaki, Benjamin Dierre and Takashi Sekiguchi
page 49

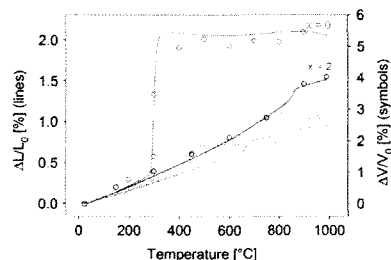


Excitation and emission spectra of $(Sr_{0.7}Eu_{0.3})Si_9Al_{19}ON_{31}$

$Sr_{0.7}Eu_{0.3}Si_9Al_{19}ON_{31}$ gives blue-emitting at 471 nm. Excitation spectrum is composed of five bands in the range of 250–450 nm, which are matching well with emission wavelength of UV LEDs.

$BaZn_2Si_2O_7$ and the solid solution series $BaZn_{2-x}Co_xSi_2O_7$ ($0 < x \leq 2$) as high temperature seals for solid oxide fuel cells studied by high-temperature X-ray diffraction and dilatometry

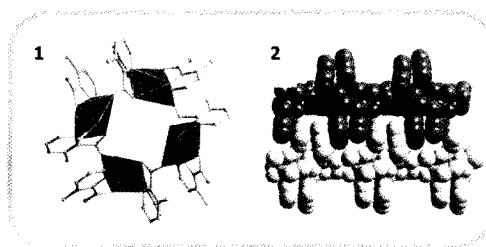
Marita Kerstan, Christian Thieme, Matthias Grosch, Matthias Müller and Christian Rüssel
page 55



The composition of the solid solution $BaZn_{2-x}Co_xSi_2O_7$ strongly affects the thermal expansion.

1D coordination polymers formed by tetranuclear lead(II) building blocks with carboxylate ligands: In situ isomerization of itaconic acid

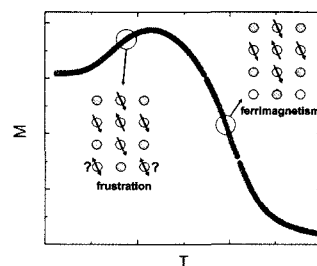
Abhinandan Rana, Swapan Kumar Jana, Sayanti Datta, Raymond J. Butcher, Ennio Zangrando and Sudipta Dalai
page 61



Two new topologically different 1D coordination polymers formed by Pb_4 clusters have been synthesized and characterized by X-ray analysis. The luminescence and thermal properties have been studied.

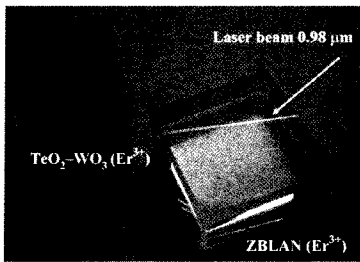
Synthesis and structural and magnetic characterization of the frustrated magnetic system $La_2Ni_{4/3-x}Co_xSb_{2/3}O_6$

D.G. Franco, R.E. Carbonio and G. Nieva
page 69



Luminescence of erbium ions in tellurite glasses

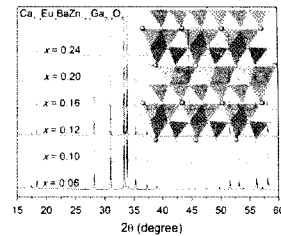
Alexander P. Savikin, Igor A. Grishin, Valery V. Sharkov and Andrei V. Budruev
 page 80



In contrast to the case of ZBLAN glass the TeO₂-WO₃ (Er³⁺) glass has bright intensity of luminescence at 1.53 μm for erbium ions that should be caused by excitation at 975 nm. Experimental data obtained have shown the possibility to use the studied glasses doped by Er³⁺ and Yb³⁺ as active elements for fiber and integrated optics.

A new member of “114” family Ca_{1-x}Eu_xBaZn_{2+x}Ga_{2-x}O₇ (x ≤ 0.24): Structure and luminescence

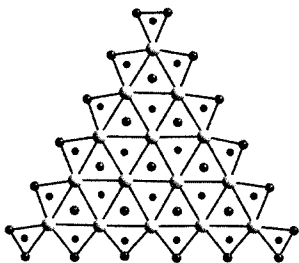
Pengfei Jiang, Zexi Liu, Xiaorui Sun, Wenliang Gao, Rihong Cong and Tao Yang
 page 105



Reitveld refinements show Ca_{1-x}Eu_xBaZn_{2+x}Ga_{2-x}O₇ all crystallize in *P6₃mc*. By Eu³⁺-doping, O3 show the most obvious movement (~0.08 Å) along the *c*-axis.

Interaction of yttrium with nickel and phosphorus: Phase diagram and structural chemistry

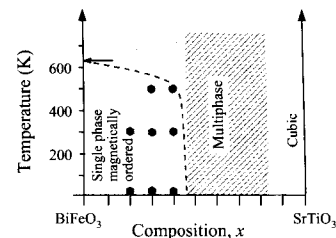
Olga Zhak, Stanislav Stoyko, Volodymyr Babizhetskyy, Olena Shved, Stepan Oryshchyn and Constantin Hoch
 page 87



Structure block with the composition Ln₁₀M₂₁X₁₅ of the homologous series of the compounds (Ln,M)₋₂X.

Crystal and magnetic structure of (1-x)BiFeO₃-xSrTiO₃ (x=0.2, 0.3, 0.4 and 0.8)

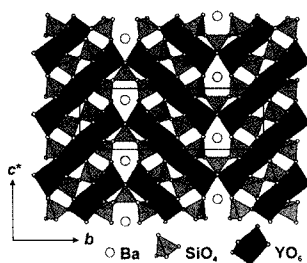
D.J. Goossens, C.J. Weekes, Maxim Avdeev and W.D. Hutchison
 page 111



A simple phase diagram for (1-x)BiFeO₃-xSrTiO₃, where hexagons indicate the rhombohedral *R3c* phase and boxes the cubic phase. Filled symbols indicate magnetic ordering. The black arrow indicates *T_N* for BiFeO₃.

Simultaneous presence of (Si₃O₁₀)⁸⁻ and (Si₂O₇)⁶⁻ groups in new synthetic mixed sorosilicates: BaY₄(Si₂O₇)(Si₃O₁₀) and isotopic compounds, studied by single-crystal X-ray diffraction, Raman spectroscopy and DFT calculations

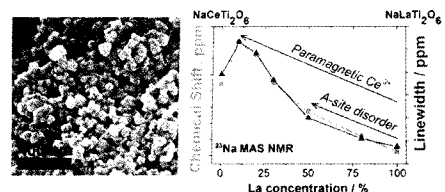
Maria Wierzbicka-Wieczorek, Daniel M. Töbrens, Uwe Kolitsch and Ekkehart Tillmanns
 page 94



View of BaY₄(Si₂O₇)(Si₃O₁₀) along [100], showing zigzag chains and the tri- and disilicate groups. The unit cell is outlined.

Investigation of the hydrothermal crystallisation of the perovskite solid solution NaCe_{1-x}La_xTi₂O₆ and its defect chemistry

Mohammad H. Harunsani, David I. Woodward, Martin D. Peel, Sharon E. Ashbrook and Richard I. Walton
 page 117



A multi-element A-site perovskite crystallises directly from aqueous, basic solutions at 240 °C; while the paramagnetic effect of Ce³⁺ on the ²³Na NMR shows a homogeneous solid-solution, the incorporation of A-site water is also found from ²H NMR and IR, with oxidation of some cerium to charge balance proved by XANES spectroscopy.

Synthesis, magnetic and electric transport properties of mixed-valence manganites $\text{La}_{0.5+x}\text{Sr}_{1.5-x}\text{Mn}_{0.5}\text{Cr}_{0.5}\text{O}_4$ ($x=0.1, 0.2$ and 0.3)

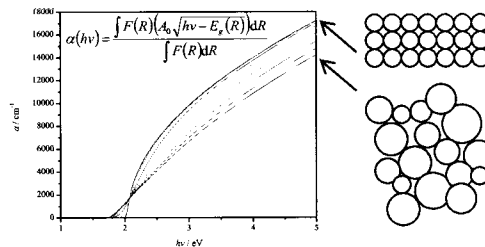
Devinder Singh and Arun Mahajan
page 126



Unit cell structure of $\text{La}_{0.6}\text{Sr}_{1.4}\text{Mn}_{0.5}\text{Cr}_{0.5}\text{O}_4$.

Three-dimensional assemblies built up by quantum dots in size-quantization regime: Band gap shifts due to size-distribution of cadmium selenide nanoparticles

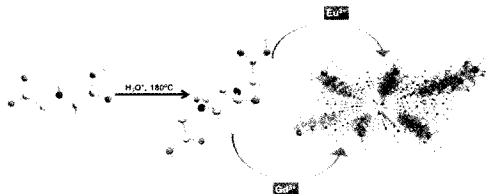
Biljana Pejova
page 147



Band gap shifts due to size-distribution of nanoparticles in 3D assemblies built up by quantum dots in size-quantization regime.

Hydrothermal reactions: From the synthesis of ligand to new lanthanide 3D-coordination polymers

Fauston Fred da Silva, Carlos Alberto Fernandes de Oliveira, Eduardo Henrique Lago Falcão, Claudia Cristina Gatto, Nivan Bezerra da Costa Jr., Ricardo Oliveira Freire, Jarosław Chojnacki and Severino Alves Júnior
page 132



Scheme of obtaining the ligand 2,5-piperazinedione-1,4-diacetic acid (H_2PDA) and two new isostructural 3D-coordination polymers $[\text{Ln}(\text{PDA})_{1.5}(\text{H}_2\text{O})](\text{H}_2\text{O})_3$ ($\text{Ln}=\text{Gd}^{3+}$ and Eu^{3+}) by hydrothermal synthesis.

New iodocuprates(I) with N-heterocyclic molecules as the cations

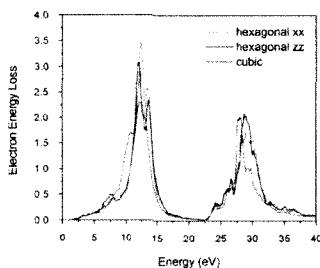
Jin-Jing Zhao, Xiao Zhang, Yan-Ning Wang, Hong-Li Jia, Jie-Hui Yu and Ji-Qing Xu
page 152



By employing hydrothermal *in situ* N-alkylation of bp/bpp with CH_3OH , three new organically templated iodocuprates(I) were obtained.

Ab initio density functional theory investigation of the structural, electronic and optical properties of Ca_3Sb_2 in hexagonal and cubic phases

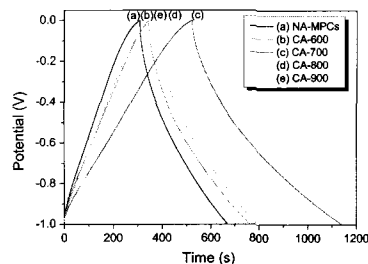
Borhan Arghavani Nia, Matin Sedighi, Masoud Shahrokhi and Rostam Moradian
page 140



A density functional theory study of structural, electronic and optical properties of Ca_3Sb_2 compound in hexagonal and cubic phases is presented.

Effects of CO_2 activation on electrochemical performance of microporous carbons derived from poly(vinylidene fluoride)

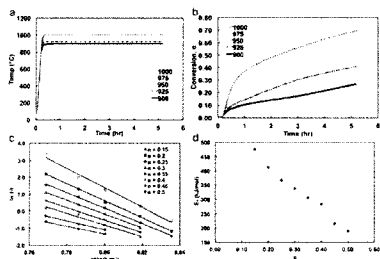
Seul-Yi Lee and Soo-Jin Park
page 158



The A-MPC samples with high specific surface area (ranging from 1030 to 1082 m^2/g), corresponding to micropore sizes of 0.67 and 0.72 nm, and with the amount of oxygen-containing groups ranging from 3.2% to 4.4% have been evaluated as electrodes for EDLC applications.

Oxidation kinetics of aluminum diboride

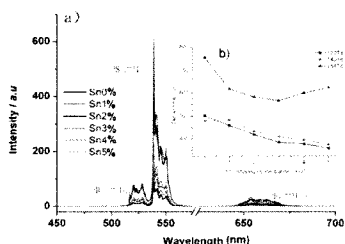
Michael L. Whittaker, H.Y. Sohn and Raymond A. Cutler
page 163



Isothermal kinetic data for AlB_2 in air, showing a constantly decreasing activation energy with increasing conversion. Model-free analysis allowed for the calculation of global kinetic parameters despite many simultaneous mechanisms occurring concurrently. (a) Time-temperature plots, (b) conversion as a function of time, (c) Arrhenius plots used to calculate activation energies, and (d) activation energy as a function of conversion.

Upconversion performance improvement of $\text{NaYF}_4:\text{Yb}$, Er by Sn codoping: Enhanced emission intensity and reduced decay time

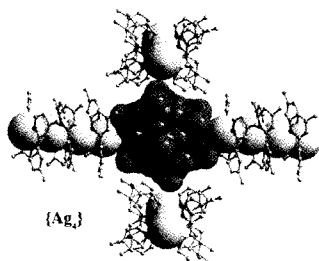
Han Yu, Wenbing Cao, Qingming Huang, En Ma, Xinqi Zhang and Jianchang Yu
page 170



Variable-valenced Sn is introduced with Yb/Er into NaYF_4 to tune structure and local crystal field. Upconversion emission intensity of Er^{3+} was enhanced while decay time constant was decreased.

Two new polyoxometalate-based hybrids consisting of Keggin-type cluster modified by $\{\text{Ag}_4\}$ group

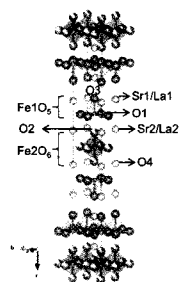
Xiaofang Zhao, Xiaowei Sun, Zhangang Han, Chuan Zhao, Haitao Yu and Xueliang Zhai
page 178



New polyoxometalate-based hybrids consisting of Keggin-type clusters modified by $\{\text{Ag}_4\}$ groups had been synthesized and characterized, and their photoluminescence properties were also discussed.

Distribution change of oxygen vacancies in layered perovskite type $(\text{Sr}, \text{La})_{n+1}\text{Fe}_n\text{O}_{3n+1}$ ($n=3$)

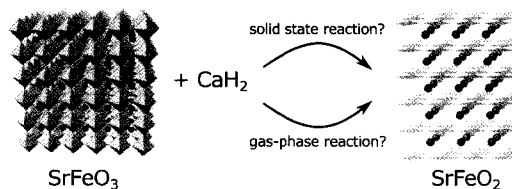
Isao Kagomiya, Keigo Jimbo and Ken-ichi Kakimoto
page 184



Crystal structure of $(\text{Sr}_{0.775}\text{La}_{0.225})_4\text{Fe}_3\text{O}_{10-\delta}$ (SLF4310) at 1000 °C.

Gas phase contributions to topochemical hydride reduction reactions

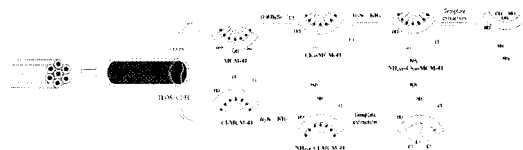
Yoji Kobayashi, Zhaofei Li, Kei Hirai, Cédric Tassel, François Loyer, Noriya Ichikawa, Naoyuki Abe, Takafumi Yamamoto, Yuichi Shimakawa, Kazuyoshi Yoshimura, Mikiyo Takano, Olivier J. Hernandez and Hiroshi Kageyama
page 190



Topochemical reductions with hydrides: Solid state or gas phase reaction?

Sol-gel synthesis of MCM-41 silicas and selective vapor-phase modification of their surface

N.V. Roik and L.A. Belyakova
page 194

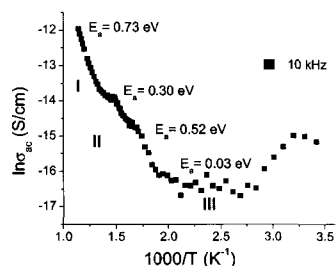


Sol-gel synthesis and postsynthetic chemical modification of template-filled MCM-41 and Cl-MCM-41 with (3-chloropropyl)triethoxysilane and 1,2-ethylenediamine in vapor phase.

Phase transition and conduction mechanism in $\text{Pb}_2\text{Na}_{0.8}\text{R}_{0.2}\text{Nb}_{4.8}\text{Fe}_{0.2}\text{O}_{15}$ material (R =rare earth)

M. Bouziane, M. Taibi and A. Boukhari

page 203

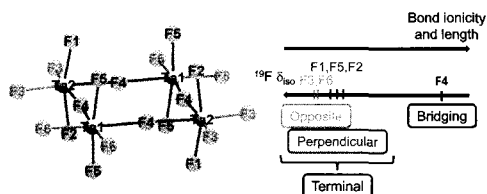


Thermal evolution of $\ln \sigma_{ac}$ of $\text{Pb}_2\text{Na}_{0.8}\text{Eu}_{0.2}\text{Nb}_{4.8}\text{Fe}_{0.2}\text{O}_{15}$ at selected frequencies.

NbF_5 and TaF_5 : Assignment of ^{19}F NMR resonances and chemical bond analysis from GIPAW calculations

Mamata Biswal, Monique Body, Christophe Legein, Aymeric Sadoc and Florent Boucher

page 208

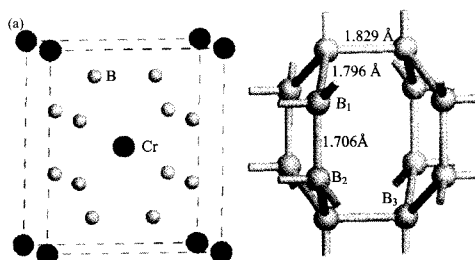


The complete assignment of the ^{19}F NMR lines of NbF_5 and TaF_5 allow establishing relationships between the ^{19}F δ_{iso} values, the nature of the fluorine atoms (bridging or terminal), the position of the terminal ones (opposite or perpendicular to the bridging ones), the fluorine charges, the ionicity and the length of the M–F bonds.

Rapid Communications

First-principles study on the structure, elastic properties, hardness and electronic structure of TMB_4 ($\text{TM}=\text{Cr}$, Re , Ru and Os) compounds

Y. Pan, W.T. Zheng, W.M. Guan, K.H. Zhang and X.F. Fan
page 29



The first-principles calculations show that the intrinsic hardness of CrB_4 and ReB_4 are bigger than 40 GPa, which are the potential superhard materials due to the B–B bonds cage structure.

Language services. Authors who require information about language editing and copyediting services pre- and post-submission please visit <http://www.elsevier.com/locate/languagepolishing> or our customer support site at <http://epsupport.elsevier.com>. Please note Elsevier neither endorses nor takes responsibility for any products, goods or services offered by outside vendors through our services or in any advertising. For more information please refer to our Terms & Conditions <http://www.elsevier.com/termsandconditions>

For a full and complete Guide for Authors, please go to: <http://www.elsevier.com/locate/jssc>

Journal of Solid State Chemistry has no page charges.